

Department of Technology Second Year B. Tech.

Scheme of Teaching and Examination Semester- III (Food Technology) To be implemented from Academic year 2021-22

Course	Course Title	Те		Credi	eme with ts Veek)	Examination Scheme (Marks)						
Code	Course Title	_	_		Total	Theory			Practical			
		L	Т	P	Credits	Scheme	Max. marks	Min. Passing	Scheme	Max. marks	Min. Passing	
FT211	Principles of Food Preservation	04	-	-	04	CIE SEE	30 70	40	-	-	-	
FT212	Engineering Mathematics-III	03	1	-	04	CIE SEE	30 70	40	IOE	50	20	
FT213	Food Microbiology	04	_	_	04	CIE	30	40	-	-	-	
					•	SEE	70	40	-	-	-	
FT214	Food Chemistry	04	-	-	04	CIE SEE	30 70	40	-	-	-	
FT215	Heat Transfer	03	_	_	03	CIE	30	40	1	-	-	
		03		_	03	SEE	70	40	-	-	-	
FT211L	Principles of food preservation Laboratory	-	-	02	01	-	-	-	EPE	50	20	
FT212L	Food Microbiology Laboratory	-	-	02	01	-	-	-	EPE	50	20	
FT213L	Food Chemistry Laboratory	-	-	02	01	-	-	-	EPE	50	20	
FT214L	Heat Transfer Laboratory	-	-	02	01	-	-	-	IOE	50	20	
FT215L	Programming Practices for Food Technologists	1	-	02	02	-	-	-	IOE	50	20	
	Total	19	1	10	25	-	500	-	-	300	-	
			•									
HS211	Environmental Studies	02	-	-	-	Project* Theory*	30 70	40	-	-	-	
	Audit Course-I											
HS212	Soft Skills Development	02	-	-	-	Institute Level	-	-	-	-	-	

Total contact hours per week: 30+2+2=34, Total Credits=25

\$ In theory student should appear for the CIE (Mid Semester Exam), submit the assignment and must secure 40% marks in SEE.

CIE: Continuous Internal Evaluation SEE: Semester End Examination
IPE: Internal Practical Evaluation EPE: External Practical Examination
IOE: Internal Oral Evaluation EOE: External Oral Examination

Note: Tutorials and Practical to be conducted in batches with batch strength not exceeding 15 students.

^{*} indicates Environmental Studies project evaluation and the theory examination will be at the end of the year i.e. along with Semester IV End Examination.



Department of Technology Second Year B. Tech.

Scheme of Teaching and Examination Semester- IV (Food Technology) To be implemented from Academic year 2021-22

Course	Company Title	,	Wit	ing So th Cre urs/W			ination e(Marks)					
Code	Course Title				Total		Theory		Practical			
		L	Т	P	Credits	Scheme	Max. marks	Min. Passing	Scheme	Max. marks	Min. Passing	
FT221	Food Process Engineering-I	03	1	_	04	CIE	30	40	-	-	-	
	Toda Trocess Engineering 1	03	•		04	SEE	70	40	-	-	-	
FT222	Food Additives and	04			04	CIE	30	40	-	-	-	
	Contaminants		-	-		SEE	70	40	-	-	-	
FT223	Food Biochemistry	04	_	_	04	CIE	30	40	-	-	-	
	rood brothermstry	0.			01	SEE	70	40	-	-	-	
FT224	Human Nutrition	04	_	_	04	CIE	30	40	-	-	-	
	Tramair Watertion	0-1			04	SEE	70	40	-	-	-	
FT225	25 Fluid Mechanics		_		04	CIE	30	40	-	-	-	
	Traid Wiceffames	04				SEE	70	40	-	-	-	
FT221L	Food Process Engineering-I Laboratory	-	-	02	01	-	-	-	IOE EOE	50 50	20	
FT222L	Food Additives and Contaminants Laboratory	-	-	02	01	-	-	-	EPE	50	20	
FT223L	Food Biochemistry Laboratory	-	-	02	01	-	-	-	EPE	50	20	
FT224L	Human Nutrition Laboratory	-	1	02	01	-	1	-	IOE	50	20	
FT225L	Fluid Mechanics Laboratory	-	-	02	01	-	-	-	IOE	50	20	
	Total	19	1	10	25	-	500	-	-	300	-	
HS221	Environmental Studies	02	i	-	-	Project* Theory*	30 70	40	-	-	-	
		_										
HS222	Introduction to Performing Arts	02	-	-	-	Institute Level	-	-	-	-	-	

Total contact hours per week: **30+2+2=34**, Total Credits=25

\$ In theory student should appear for the CIE (Mid Semester Exam), submit the assignment and must secure 40% marks in SEE.

CIE: Continuous Internal Evaluation SEE: Semester End Examination

IPE: Internal Practical Evaluation EPE: External Practical Examination

IOE: Internal Oral Evaluation EOE: External Oral Examination

Note: Tutorials and Practical to be conducted in batches with batch strength not exceeding 15students.

^{*} indicates Environmental Studies project evaluation and the theory examination will be at the end of the year i.e. along with Semester IV End Examination.

Class, Part & Semester	:	Second	Second Year B. Tech. (Food Technology), Part II, Semester III							
Course Title	•	Principl	Course Code:	:	FT211					
Teaching Scheme		Lecture : 4 4 hours/week			Total		0.4			
(Hours)	•	Tutorial : Nil	Nil	Credits	•	04				
Evaluation Scheme		CIE=30	SEE - 70	Grand Total=	Duration	:	3 hours			
(Marks)	•	(20+10)	SEE = 70	100	of SEE		5 110015			
Revision:	:	Fourth			Month	:	June 2021			
Pre-requisites	:	knowledge of	basic science							
(if any)										
Course Domain	:	Core								

A Principle of Food Preservation is a multi disciplinary course aimed towards imparting knowledge for students studying Food Technology. The course has wide applications in food processing industry (fruits and vegetables processing, bakery processing, and all other ready to eat packed foods and beverages, etc.). The course is completely Industry oriented which includes all the techniques that are expected in commercial industry. Hence the student will be well versed with advanced technique which is need of the hour for providing employment opportunities.

	- r - 0 - r - 7							
	Course Objectives		Course Outcomes					
1.	The course teacher will support students	1.	Students will be able to understand the					
	in applying basic food science knowledge		need for food processing.					
	and understanding of biochemical changes							
	that occur during various processing and							
	conservation techniques.							
2.	The course teacher will introduce students	2.	Students will be able to grasp the various					
	to different food processing techniques.		food processing techniques.					
3.	The course teacher will educate students	3.	Students will be able to understand the					
	on the technical mechanism for preserving		different preservation technique.					
	food.							
4.	The course teacher will introduce students	4.	Students will be able to understand the					
	to food preservation methods to avoid		principles of food spoilage and the ways					
	waste.		to prevent.					
5.	The course teacher will introduce students	5.	Students will be able to describe the					
	to different non thermal ways of food		principles involved in non thermal food					
	processing.		processing.					
6.	The course teacher will introduce students	6.	Students will be able to describe the					
	to different modern ways of processing		principles involved in the various modern					
	and conserving food.		ways of food processing.					

	Curriculum Content	Hours
Unit I	Introduction	
Scope	and Importance of food processing, National and International perspectives,	8
Objec	tives and techniques of food preservation.	
Unit I	Food preservation by low temperature	8
Coolir	ng, Refrigeration, freezing and freeze drying	0
Unit I	II Food preservation by heating	8
Blancl	ning, pasteurization, sterilization, UHT processing, extrusion cooking of food.	8
Unit I	V Preservation by drying	
Proce	ssing and preservation by drying, concentration and evaporation-types of dryers and	9
their	suitability for different food products;	
Unit \	Food preservation by Non-thermal method	
Chem	ical preservation, fermentation methods for food preservation, irradiation, membrane	9
techn		
Unit \	/I Recent methods for food preservation	
	d electric field processing, high pressure processing, processing by using ultrasound,	10
	tric, ohmic and infrared heating etc.	
	sted list of Tutorials and Assignments:	
1) Ma	rket Survey	
Sugge	ested Text Books:	
1.	Fellows, P. and Ellis H. (1990). "Food processing technology: principles and practice"	", Wood
1.	Head Publishing Ltd.	
2.	Manoranjan Kalia and Sangita Sood. (2019). "Food preservation and processing",	Kalyani
۷.	Publishers. New Delhi.	
3.	Richardson, T. and Finley, J.W. (2003). "Chemical changes in food during produced	cessing".
J.	Macmillon Publishers. Canada.	
4.	Desrosier, N.W. (2018). The Technology of Food Preservation.3 rd Edition. Medtech Pub	olishers.
Sugge	ested Reference Books:	
1.	Jelen, P. (1985). Introduction to Food Processing. Prentice Hall, Reston Virginia, USA.	
2.	Heldman, D.R. and Singh R. P. (2016). Introduction to Food Engineering.5 th Edition.	Elsevier
	India	
3.	William C. Frazier And Dennis C. Westoff (2017). Food Microbiology 5th Edition, McG	Graw Hill
	Education.	
4.	Singh, Anju. (2017). Handbook of Food Preservation. Agrotech Press.	
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Class, Part & Semester	:	Second	Second Year B. Tech. (Food Technology), Part II, Semester III							
Course Title	••	Engine	Course Code:	:	FT212					
Teaching Scheme (Hours)		Lecture : 3	Lecture : 3 03 hours/week				0.4			
		Tutorial : 1	01 hour/wee	Credits	•	04				
Evaluation Scheme		CIE=30	SEE = 70	Grand Total=	Duration	:	3 hours			
(Marks)	•	(20+10)	SEE = 70	100	of SEE					
Revision:	••	Fourth			Month	:	June 2021			
Pre-requisites	••	BS-11A1, BS-1	3S-11A1, BS-12A1							
(if any)										
Course Domain	:	Basic Science	sic Science							

This course is about the basic mathematics that is fundamental and essential component in all streams of undergraduate studies in sciences and engineering. The course consists of topics in differential equations, partial differential equations, Laplace transform - its inverse, Vector calculus and probability with applications to various engineering problems.

	Course Objectives		Course Outcomes
1.	The course teacher will describe solution of LDE and its applications.	1.	Students will be able to solve Linear Differential Equations and apply them to realistic problems.
2.	The course teacher will introduce Laplace Transform.	2.	Students will be able to gain the basic knowledge of Laplace transform and their applicability in solving initial value problems.
3.	The course teacher will introduce Partial Differential Equations and its Applications.	3.	Students will be able to solve Partial Differential Equations for solving problems in Food Engineering.
4.	The course teacher will introduce Probability and its Applications.	4.	Students will be able to solve engineering problems using Probability.
5.	The course teacher will explain Vector Differentiation and Vector Integration.	5.	Students will be able to analyze and solve engineering problems using vector differentiation.
6.	The course teacher will help to formulate a mathematical model of a real life and engineering problem, solve and interpret the solution in real world.	6.	The student must be able to formulate a mathematical model of a real life and engineering problem, solve and interpret the solution in real world.

Curriculum Content	Hours
Unit: I Linear Differential Equations:	07

Linear Differential Equations with constant coefficients, Homogenous Linear differential					
equations, method of variation of parameters					
Unit: II Laplace Transform:					
Definition, properties of Laplace transforms, transforms of derivatives, transforms	07				
ofintegral, Inverse Laplace transforms, Convolution theorem. Applications to initial value	07				
boundary problems, Heaviside Unit step function, Dirac delta function, Periodic function.					
Unit: III Partial differential equations:	06				
Four standard forms of partial differential equations of first order.					
Unit: IV Application of Partial differential Equations					
Classification of PDE, Solution of Wave Equation, One dimensional heat equation and two	06				
dimensional Laplace equations by the method of separation of variables, use of Fourier					
series.					
Unit: V Probability:					
Random variable, Probability mass function and probability density function, Binomial,	06				
Poisson and Normal distributions.					
Unit: VI Vector Calculus:					
Vector Differentiation: Differentiation of vectors, Gradient of scalar point function,					
Directional derivative, Divergence of vector point function, Curl of a vector point function.	07				
Irrotational and solenoidal vector field.	07				
Vector Integration: The line integral, Surface integral, volume integral, Gauss's Divergence					
theorem, Stoke's theorem, Green's theorem (Without proof).					
Suggested list of Tutorials and Assignments:					

Suggested list of Tutorials and Assignments:

- 1. To find solution of LDE with constant coefficients
- 2. Applications of LDE
- 3. Laplace Transform
- 4. Partial Differential Equations
- 5. Applications Of PDE
- 6. Probability
- 7. Vector differentiation
- 8. Vector Integration

General Instructions:

- 1. Batch wise tutorials are to be conducted. The number of students per batch should be as per the practical batches.
- 2. Each Student has to write at least 6 assignments on entire syllabus.

Suggested Text Books:

1.	Kreyszig, Erwin. (2015).Advanced Engineering Mathematics. 10 th Edition.Wiley India Pvt. Ltd.
2.	Grewal,B. S. (2014). Higher Engineering Mathematic. Khanna Publications, New Delhi.
3.	Ramana, B.V. (2017). Higher Engineering Mathematics", McGraw Hill Education India.
4.	Wartikar, P. N. and Wartikar, J. N., "A Text Book of Engineering Mathematics (Volume-I, II and

	II)", Pune Vidyarthi Griha Prakashan, Pune.						
5.	H. K. Das, "Advanced Engineering Mathematics", S. Chand Publication.						
Sugge	Suggested Reference Books:						
1.	Wylie, C.R. Advanced Engineering Mathematics", McGraw Hill Publication, New Delhi.						
2.	Merle C. Potter, "Advanced Engineering Mathematics", OXFORD University Press, 3 rd Edition						
3.	Shanti Narayan, "Differential Calculus" S. Chand and company, New Delhi.						
4.	Sastry, S. S. "Engineering Mathematics (Volume-I)", Prentice Hall Publication, New Delhi.						
5.	M. D. Greenberg, "Advanced Engineering Mathematics", Pearson Education.						

Class, Part & Semester	:	Second '	Second Year B. Tech. (Food Technology), Part II, Semester III						
Course Title	:	F	Course Code:	:	FT213				
Teaching Scheme		Lecture : 4	Total		04				
(Hours)	•	Tutorial : Nil	Nil		Credits	•	04		
Evaluation Scheme		CIE=30	SEE = 70	Grand Total=	Duration	:	3 hours		
(Marks)	•	(20+10)	SEE = 70	100	of SEE				
Revision:	:	Fourth	Fourth				June 2021		
Pre-requisites	:	knowledge of	basic science						
(if any)									
Course Domain	:	Core							

Food microbiology is the study of the microorganisms that inhabit, create, or contaminate food. Microbes are being used for the production of foods, food supplements and food ingredients. Microorganisms are widely used in fermentative production, processing, and preservation of various kinds of foods and beverages. Microorganisms are also responsible for food spoilage and food poisoning which result in loss of food products and also cause damage to human and animal health causing health and economic losses. Studying the food microbiology and the effects of microbial contamination is vital to various food safety, production, processing, preservation, and storage aspects.

	Course Objectives		Course Outcomes
1.	The course teacher will support students in	1.	Students will be able to understand the
	understanding the various microbes		various microbes associated with food
	associated with food and food groups.		and food groups.
2.	The course teacher will familiarize the role	2.	Students will be able to familiarize the
	of microbes in food spoilage and food		role of microbes in food spoilage and
	preservation.		food preservation.
3.	The course teacher will educate students to	3.	Students will be able to detect
	detect pathogens in food.		pathogens in food.
4.	The course teacher will describe the	4.	Students will be able to describe the
	beneficial role of microorganisms in		beneficial role of microorganisms in
	fermented foods and in food processing.		fermented foods and in food
			processing.
5.	The course teacher will introduce students	5.	Students will be able to apply the
	to different non thermal ways of food		theories and principles of food
	processing.		microbiology in practical, real-world
			situations and problems.
6.	The course teacher will introduce students	6.	Students will be able utilize laboratory
l			1

	to utilize laboratory techniques to detect, quantify, and identify microorganisms in foods. techniques to detect, quantify, and identify microorganisms in foods.								
	Curriculum Con	ntent		Hour s					
	Introduction to microbiology ation and Scope of microbiology. General M	licrobiolo	ngy Cultural characteristics and						
reproduction of bacteria, yeasts, fungi, actinomycetes, algae, protozoa and rickettsia. Genetic recombination, transduction, transformation and bacterial conjugation, mutation and mutagenesis.									
	II Growth curve all and Chemical factors affecting growth and	l destruc	tion of micro-organisms, growth	8					
	III Viruses	f	h	8					
Structure and reproduction with particular reference to food born viruses. UNIT IV Food Preservation									
Principles of Food Preservation, Methods of preservations, Drying, Dehydration, Freezing, Chemical Preservation, Mechanical destruction and Maintenance of anaerobic condition.									
UNIT V Spoilage of Food Microbial Spoilage of cereals and cereal products, milk and milk products, fruit and Vegetable products, meat, poultry egg and fish products, sugar and Sugar Products.									
	VI Food poisoning cation, Food borne illness			6					
Sugge	sted Text Books:								
1.	Prescott and Dunn. (2004). Industrial m distributors Delhi.	nicrobiolo	ogy, 4th edition, CBS publisher	s and					
2.	2. Purohit S. S. (2003). Microbiology fundamentals and applications" Edition, 6. Publishe Agrobios.								
3. William C. Frazier And Dennis C. Westoff. (2017). Food Microbiology 5th Edition, McGrav									
Sugge	sted Reference Books:								
1.	Jay, J. M. (2005). Modern Food Microbiology.	4th ed. (Chapman & Hall. New York, N.Y.						
2.	Mossel, D.A.A., Corry, J. E. L., Struijk, C. Microbiology of Foods. John Wiley & Sons. Ne		, ,	of the					

Class, Part & Semester	:	Second	Second Year B. Tech. (Food Technology), Part II, Semester III					
Course Title	•	F	Course Code:	••	FT 214			
Teaching Scheme		Lecture : 4	4 hours/week	Total		04		
(Hours)	•	Tutorial : Nil	Nil		Nil Nil	Credits	•	04
Evaluation Scheme		CIE=30	SEE = 70	Grand Total=	Duration	:	3 hours	
(Marks)	•	(20+10)	3LL = 70	100	of SEE			
Revision:	:	Fourth	Fourth Month : June 202			June 2021		
Pre-requisites	:	Knowledge of Engineering Chemistry						
(if any)								
Course Domain	:	Core	Core					

Course Rationale: The course covers a broad range of subjects related to foods including basic scientific principles to food systems and practical applications. Food chemistry deals with the chemical, physical and functional properties of food constituents and the chemical changes these constituents undergo during handling, processing and storage including those that limit food shelf life. Food chemistry is a major aspect of Food science, that is an interdisciplinary subject in which the engineering, biological, and physical sciences are used to study the nature of foods, the causes of deterioration, the principles underlying food processing, and the improvement of foods from a consumer and sustainability perspective. The aim of the course is to provide the students with a deep understanding of how food components contributes to overall quality of foods; and to enable students to evaluate and explain how the highly complex nature of food may result in a multitude of desired and undesired reactions which are controlled by a variety of parameters.

	Course Objectives		Course Outcomes
1.	The course teacher will help to understand functions of Food Chemistry and basic physico-chemical properties and chemical structures of food components.	1.	Students will be able to describe the various constituents present in foods and their roles respectively.
2.	The course teacher will helps to understand the importance and mechanisms of the reactions of food components taking place during food processing.	2.	Students will be able to describe the mechanisms and significance of physicochemical reactions involved in food processing and subsequent storage.
3.	The course teacher will help to understand the significance and mechanisms of the reactions of food components taking place storage and spoilage.	3.	Students will be able to report the mechanisms and significance of physicochemical reactions involved in spoilage of foods.
4.	The course teacher will help to think critically on the role of water and its	4.	Students will be able to explain the significance of water in food quality,

	various forms in food preservation.		preservation and storage.				
5.	The course teacher will help to 5. Students will be able to describe						
٥.	understand the role of food constituents	٥.	demonstrate the role of				
	responsible for quality of foods.			nal/anti-			
			nutritional and aesthetic quality	or raw			
			and processed foods.				
6.	The course teacher will help to apply	6.	Students will be able to extrapo				
	course concepts in solving problems		knowledge gained on food com	•			
related to food constituents. to practical problems in food qual							
	Curriculum Con	tont		Hours			
Linit ir	ntroduction to Food Chemistry	tent		110013			
	tance of Food Chemistry, An introduction	n to f	food resources and its general	05			
-	osition	ווי נט ו	ood resources and its general	03			
	Water						
	in food systems – Role and types of water in	foods	Concent of water activity Effect	07			
	er activity on food processing and storage	10003,	concept of water activity, Effect	07			
	I Carbohydrate			. <u></u>			
	istry of carbohydrates (Classification, Stru	ıctura	and Sources and properties)				
	rations of Carbohydrates in Food processing In		, ,	10			
	n Health.	uusti y.	concept of fiber and its effect of				
	V Protein						
	istry of amino acids and their properties (isoeld	octric n	H solubility profile). Functions of				
	ns; Peptides; classification of proteins; stru	-		10			
	rnary); Denaturation of proteins; estimation			10			
· ·	n in Food processing Industry.	i di pi	otems in 1 dous, Applications of				
	Lipids						
	istry of Lipids (Sources, Classification and prop	artias)	Processing of ail Hydrogenation				
	s, Changes occurring in fats and oils during	•	, ,	10			
	ssing Industry.	proces	ising, Applications of Fatili Food				
	/I Vitamins and Minerals			<u> </u>			
	ins: Classification of Vitamins, Sources of Vita	mins n	ronerties effect of processing on				
	ins and deficiency problems of Vitamins.	ιιιιιο, μ	roperties, effect of processing of	10			
	, ,	nortics	of minorals offort of processing	10			
Minerals : Classification, Structure, Sources and properties of minerals, effect of processing							
on of Minerals and deficiency problems of minerals Suggested Text Books:							
1.	deMan, John M.(1999).Principles of Food Ch	omictr <i>i</i>	2rdEdition Springer				
2.	Meyer, Lillian Hoagland. (1987). Food Chemis		, , <u> </u>				
		•		rk			
	3. Belitz, H.D. Grosch. Food Chemistry. 3 rd revised ed. Springer Berlin, Heidelberg, New York.						
	4. O. R. Fennema. Food Chemistry. Marcel Dekker, Inc., New York.						
Sugge	Suggested Reference Books:						

1.	Chopra, H.K. and Panesar, P.S. (2010). Food Chemistry. Narosa.
2.	Aurand, L.W. and Woods, A.E. (1973). Food Chemistry. Avi Publishing Company, Inc, Westport,
	СТ.
3.	Morris B Jacob. The Chemical analysis of foods and food products, 3 rd Edition, Vam Nostr and
	Co, Princeston, New Jersey.
4.	Sakuntala Manay and Shadaksaraswamy, M. (1987). Food facts and principles. Allied
	Publishers, New Delhi.

	Department		ecnnology, B. Tecn.	(1000 1	CCIIIIOIOE	,y, i rogrami synan	ous w. c. 1. 2021		•
Class,	Part & Semester		Second Year B. Tech. (Food Technology), Part II, Semester III						
	Course Title	:		Heat Transfer		Course Code:	:	FT215	
Tec	aching Scheme		Lecture : 3	Lecture : 3 03 hours/week					0.2
	(Hours)	:	Tutorial : Nil Nil				Credits	:	03
Eva	luation Scheme (Marks)	:	CIE=30 (20+10)	SEE = 70 Total=100		Duration of SEE	:	3 hours	
	Revision:	:	Fourth				Month	:	June 2021
P	Pre-requisites (if any)	:	Knowledge of Engineering Ma	_	_	Physics, Engi	neering Math	ien	natics I and
Co	ourse Domain	:	Core						
Course Rationale: The course is prepared to provide the detailed understating of heat transfer principles, numerical analysis, and heat exchanger and evaporation process.									
	Course O						rse Outcomes		
2.	understand mod formulate the e heat rate of cond The course teach empirical equati convection for diff The course to understand phase and latent heat of free convective, respective, resp	es qua uct er ons fer eac ese- of v	ations for calcuion. will help to derive of free or forent fluid flow. ther will help to the properties of the propert	r and lating re the orced to to mena uding	2.	mechanisms of models to so state heat con Students will parameters wheat transfer convection.	be able to ider of heat transfe plve one- dir duction proce be able to re- which affect mode either be able to boiling, conde	r a mei ss. cog the fo	nd formulate nsion steady gnize the key e convective erced or free plement the
4.	The course tea various laws to u transfer.		•		4.	Students will evaluate the between two			analyze and eat transfer
5.	The course teachers analyze heat transfer systems	nsf ecti red to s	er equipment and on of heat exchates esigning of the save energy).	nd to ngers heat	5.	exchangers us	be able to ing LMTD and	NT	'U method.
6.	The course teach types of evaporator select evaporator	ato s fo	rs further to a	ssess,	6.		be able to poration pro		•

improve their performance through heat

	transfer knowledge.					
	Curriculum Content	Hours				
Uni	t I: CONDUCTION					
Fun	damentals: Introduction to three modes of heat transfer: Conduction, convection and					
radi	ation. General laws of heat transfer.					
	duction: Fourier's law, Thermal Conductivity- its variation with temperature and Pressure	80				
	its relationship with electrical conductivity. Heat transfer through composite walls and					
	nders. Unsteady state heat transfer through some important shapes. Different types of					
insulating materials, general properties and application of insulators, Problems.						
_	t II: CONVECTION					
	vection, types of convection and relationship between individual and overall heat transfer					
	fficient, dimensional co-relation for convective heat transfer, fouling factors. Forced					
	vection: In laminar flow-Heat transfer in plate and tubes. In turbulent flow-Empirical	09				
· -	ations for individual coefficients: inside tubes, outside tubes, outside bundle of tubes,					
flow past spheres. Significance of Prandtl number, Nusselt number, Grashof number, Graetz						
number and Peclet number etc. Correction for tube length. Corrections for heating and						
	ling the fluid. Various analogies between heat & momentum transfer, Problems.					
Unit III: Heat transfer with phase change						
	ing of liquids, Pool boiling curve, different types of pool boiling, Condensation of vapor, wise and drop wise condensation, Energy balance.	06				
	t IV: Radiation					
	iation laws like Stefan Boltzmann's law, Kirchhoff's law, Wien's law, Plank's law etc. Black					
	y, Grey body. Transmissivity, Absorptivity, Reflectivity, Emissivity of black bodies and gray	04				
	ies. Application of thermal radiation for Food process industry: Radiation Transfer	04				
	ween surfaces. Radiation through semi transparent materials.					
	t V: Heat Exchange equipment					
	t exchange equipments and its classification, LMTD, LMTD correction factors and Flow					
	ingements in heat exchanger. Shell and tube heat exchanger, double pipe, plate type,	06				
	Graphite block, fin tube heat exchanger.					
Uni	t VI: Evaporation					
Eva	poration, capacity, steam economy, material and enthalpy balance for single effect	06				
eva	porators, classification and application of evaporators.					
	·					
Sug	gested Text Books:					
1.	McCabe, W. L., Smith, J. C.and Harriott, P. (2005). Unit Operations of Chemical Enginee	ring. 7 th				
	Edition. McGraw Hill.	J				
2.	Lienhard, John H. (2011). A Heat Transfer Textbook.4th Edition., Dover Publications Inc.					

3.	J. P. Holman. Heat Transfer.10 th Edition. McGraw Hill.
Sug	ggested Reference Books:
1.	McAdams, W. H. Heat Transmission.3 rd Edition. McGraw Hill.
2.	Kern, D. Q. Process Heat Transfer. 1 st Edition. McGraw Hill.
3.	Don W. Green, Robert H. Perry. Perry's Chemical Engineer's Handbook. 8 th Edition. McGraw-Hill:
	New York.
4.	Warren, M. Rohsenow, James, P. Hartnett and Young I. Cho (1998). Handbook of Heat Transfer,
	McGraw-Hill Education: New York.

Class, Part & Semester	:	Seco	Second Year B. Tech. (Food Technology), Part II, Semester III					
Course Title	:	Principle of	Food Preservati	Course Code:	:	FT211L		
Teaching Scheme (Hours)	:	Practical :-	2 hours/week		Total Credits	:	01	
Evaluation Scheme (Marks)	:	IPE: Nil IOE: Nil	EPE =50	Total = 50	Duration of EPE	:	2 hours	
Revision:	:	Fourth	Fourth			:	June 2021	
Pre-requisites (if any)	:	knowledge of basic science						
Course Domain	:	Core	ore					

This course has wide applications in food processing industry (fruits and vegetables processing, canning, dairy processing and beverages, etc.). The course is completely Industry oriented which includes all the practical techniques that are expected in commercial industry. Hence the student will be well versed with basic and advanced technique which is need of the hour for providing employment opportunities.

	Course Objectives		Course Outcomes				
1.	The course teacher imparts knowledge and expertise on preservation and food processing methods.	1.	Students will identify processing equipment and select suitable processing equipment and storage methods for certain foods.				
2.	The course teacher will help students familiarize themselves with good manufacturing practices and standard operating procedures used in laboratory activities.	2.	Students will familiarize themselves with good manufacturing practices and standard operating procedures used in lab exercises.				
3.	The course teacher will teach to preserve food through drying.	3.	Students will be able to preserve food through drying.				
4	The course teacher will teach to preserve food through freezing.	4.	Students will be able to preserve food through freezing.				
5.	The course teacher will teach to preserve food with the help of sugar, salt and acids.	5.	Students will be able to preserve food with the help of sugar, salt and acids.				
6	The course teacher will assist with the quality assessment of conservation products.	6.	Students will be able to evaluate the quality of the preserved products.				
Sr. No.	List of Experiments						
1.	Introduction to food processing equipments						
2.	To study effect of blanching on quality of foods						

3.	To check the adequacy of Blanching treatment
4.	Preservation of food by the process of freezing
5.	Drying of food using Tray dryer/other dryer
6.	Preparation of product by using sugar as preservative
7.	Preparation of product by using salt as preservative
8.	Preservation of product by using chemical preservatives
9.	Preservation of food by canning
10.	Extrusion cooking of food
11.	Food Fermentation
12.	Market Survey
Sugges	ted Text Books/ Reference Books/Manual
1.	Fellows, P. J. (2009). Food processing technology: principles and practice. Elsevier.
2.	Singh, R. P., and Heldman, D. R. (2001). Introduction to food engineering. Gulf Professional
۷.	Publishing.
3.	Desrosier, N. W., and James N. Desrosier. (1977). The technology of food preservation. 4 th Ed.
٥.	AVI Publishing Company, Inc.

Class, I	Part & Semester	:	Second Year B. Tech. (Food Technology), Part II, Semester III						
Ca	ourse Title	:	Food Microbiology Laboratory			Course Code:	:	FT212L	
	Teaching Scheme (Hours)		Practical :-	2 hours/wee	ek		Total Credits	:	01
	ation Scheme (Marks)	:	IPE: Nil IOE: Nil	EPE =50 Total = 50		Duration of EPE	:	2 hours	
F	Revision:	:	Fourth				Month	:	June 2021
Pre	r-requisites	:	knowledge	of basic scien	ce		l	1 1	
	(if any)								
Cou	rse Domain	:	Core						
	Course	Ob,	jectives			Со	urse Outcom	es	
1.	The course tead	he	r will imparts	knowledge	1.	Students	will be ab	le 1	to understand
	to understand is	sola	ation, charact	erization of		isolation,	characteriz	atio	n of various
	various microbes associated with foods			with foods		microbes	associated w	ith ·	foods and food
	and food groups.				groups.	groups.			
2.	The course te	acl	her will hel	p students	2.	Students	nts will familiarize themselves with		
	familiarize them	ıse	lves with mic	robiological		microbiol	iological techniques for the study		
	techniques for t	he	study of food	ds.		of foods.	of foods.		
3.	The course tea	ich	er will teach	n to detect	3.	Students	will be	abl	e to detect
	microorganism	in f	oods.			microorga	anisms in foo	ds.	
4	understand	tl	acher will he activi	teach to ities of	4.		will be able of microorga		understand the ns in food.
5.	microorganisms			+	5.	Ctudonto	المام مما الثنيي		
5.	The course tea the microorgani				5.		anisms presei		quantify the
6	The course to		<u> </u>		6.		<u> </u>		describe the
	characteristics				0.				borne, water
	and spoilage		microorgani						rganisms, and
	methods for th		_	-			_		ion, detection
	identification.				and ident				
				l					
Sr. No.	List of Experime	ent	s						
1.	Study of instrur	nei	nts used for n	nicrobiology,	cleani	ng and steriliza	ntion of glassy	ware	9
2.	Preparation of I	me	dia, techniqu	es of incubati	on				
3.	Staining metho	ds	(monochrom	e staining, gra	ım sta	ining, flagella s	staining, caps	ule	staining and
	endo spore stai	nin	g)						

4.	Quantification of Microbes: Sampling and Serial Dilution; Bacterial count in food products—TVC
5.	Pure culture techniques (streak plate/pour plate), growth curve
6.	Isolation of molds from foods, microbial examination of cereal and cereal products
7.	Microbial examination of fruits and vegetables
8.	Microbial examination of milk and milk products, microbial examination of sugar, salt and spices
9.	Microbial examination of meat and meat products
10.	Microbial examination of water
11.	Microbial examination of fermented food
12.	Enumeration and Isolation of Staphylococci from ready to eat street foods
13.	Effect of cleaning and disinfection on microbial load effect of cleaning and disinfection on microbial load
Genera	Instructions:-
Suggest	ted Text Books/ Reference Books/Manual
1.	Frazier, W.C., and Westhoff, D.C. 1988. Food Microbiology, 4 th ed. McGraw-Hill, New York.
2.	Mossel, D. A. A., Corry, J. E. L., Struijk, C. B., and Baird, R. M. (1995). Essentials of the Microbiology of Foods. John Wiley & Sons. New York, NY
3.	Purohit S. S. (2003). "Microbiology fundamentals and applications" Edition, 6. Publisher, Agrobios.

Class, Part & Semester	:	Second Year B. Tech. (Food Technology), Part II, Semester III						
Course Title	:	Food Chemistry Laboratory			Course Code:	:	FT 213L	
Teaching Scheme (Hours)	:	Practical :	2 hours /week		Total Credits	••	01	
Evaluation Scheme (Marks)	:	IPE: Nil IOE: Nil	EPE: 50 EOE: Nil	Total= 50	Duration of EPE	:	2 hours	
Revision:	:	Fourth	Fourth				June 2021	
Pre-requisites (if any)	:	Knowledge of Engineering Chemistry						
Course Domain	:	Core	Core					

Course Rationale: The course covers a broad range of subjects related to foods including basic scientific principles to food systems and practical applications. Food chemistry deals with the chemical, physical and functional properties of food constituents and the chemical changes these constituents undergo during handling, processing and storage including those that limit food shelf life. Food chemistry is a major aspect of Food science, that is an interdisciplinary subject in which the engineering, biological, and physical sciences are used to study the nature of foods, the causes of deterioration, the principles underlying food processing, and the improvement of foods from a consumer and sustainability perspective. The aim of the course is to provide the students with a deep understanding of how food components contributes to overall quality of foods; and to enable students to evaluate and explain how the highly complex nature of food may result in a multitude of desired and undesired reactions which are controlled by a variety of parameters.

	Course Objectives		Course Outcome				
1.	The course teacher will train the students with hands on experience with chemical compositions of foods.	1.	Students will be able to understand the principles behind analytical techniques associated with food.				
2.	The course teacher will assist them in analysis of various food constituents, additives present in the food.	2.	Students will be able to select the appropriate analytical technique when presented with a practical problem.				
3.	The course teacher will help to study the basic nutrients and their requirements for human nutrition.	3.	Students will be able to demonstrate practical proficiency in a food analysis laboratory.				
4.	The course teacher will help to evaluate new product development.	4.	Students will be able to describe and use principal analytical methods used for quantifying the composition and reactions of food components.				
Sr. No.	List of Experiments						
1.	Determination of moisture by hot air oven	method					

2.	Estimation of crude protein by micro-Kjeldahlmethod
3.	Estimation of gluten
4.	Estimation of ash / mineral matter
5.	Estimation of crude fat
6.	Determination of acid value of oil
7.	Determination of saponification value/number of an oil or fat
8.	Determination of reducing sugar and total sugar
9.	Estimation of starch by Anthrone reagent
10.	Determination of vitamin c content
11.	Extraction of chlorophyll from leaf tissue
12.	Extraction of carotenoids from leaf tissue
Suggest	ed Text Books/ Reference Books/Manual
1	Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to
1.	follow.
2.	Association of Official Agricultural Chemists, and Horwitz, W. (1975). Official methods of
۷.	analysis (Vol. 222). Washington, DC: Association of Official Analytical Chemists.

Class, Part & Semester	:	Second Year B. Tech. (Food Technology), Part II, Semester III					
Course Title	:	Heat	Heat Transfer Laboratory				FT214L
	•	1100	ricat fransier Laboratory		Code:	Ī	
Teaching Scheme	:	Dractical :	Practical : 2 hours/week		Total		01
(Hours)		Practical			Credits	•	01
Evaluation Scheme		IPE= Nil	EPE= Nil	Total=50	Duration		2 hours
(Marks)	•	IOE=50	EOE=Nil		of EPE		2 110u13
Revision:	:	Fourth			Month	:	June 2021
Pre-requisites	:	Knowledge o	of Engineering	Physics, Engi	neering Ma	the	matics I and
(if any)		Engineering N	Nathematics II				
Course Domain	:	Core					

The main purpose to apply the basic of heat transfer. It also provides an idea about various equipments used for heat transfer.

	Course Objectives	Course Outcomes			
1.	The course teacher will recognize food process for fundamentals of some major heat transfer operations through experimental set ups.	1.	Students will be able to understanding fundamentals of some major Heat transfer operations through practical work.		
2.	The course teacher will explain them hands on training on important heat transfer devices and to motivate them for team work	2.	Students will be able to work in group and imitate Standard Operating Procedure for practical work v.i.z. responding to the guide.		
3.	The course teacher will help to apply new ideas in a miniature process plant environment using steam and the other process utilities.	3.	Students will be able to learn design principles and be mechanic for heat transfer devices.		
4.	The course teacher will demonstrate the skill of learners in safe handling of major heat transfer equipment/devices, in close observation of their operation, in developing analytical ability in correlating the performance of the devices with their operational conditions	4.	Students will be able to build foundation for process intensification and be adapted to handle heat transfer operations.		

Sr. No.	ist of Experiments					
1.	Thermal conductivity of metal rod apparatus					
2.	Composite wall apparatus					
3.	Thermal conductivity of insulating powder apparatus					

4.	Lagged pipe apparatus						
5.	Heat transfer by natural convection apparatus						
6.	Heat transfer by forced convection apparatus						
7.	Study of two phase flow apparatus						
8.	Stefan Boltzmann apparatus						
9.	Emissivity measurement of radiating surfaces apparatus						
10.	Parallel flow/counter flow heat exchanger apparatus						
Genera	Instructions:						
Suggest	red Text Books/ Reference Books/Manual						
1.	Kern, D. Q. Process Heat Transfer. McGraw-Hill, 1st Edition.						
2.	Binay K. Dutta .Heat Transfer: Principles and Applications, Prentice-Hall of India, 1st Edition.						
3.	William H. Mc Adams .Heat Transmission. McGraw-Hill, 3 rd Edition.						

Class, Part & Semester	:	Second Year B. Tech. (Food Technology), Part II, Semester III					
Course Title	:	Programming Practices for Food Technologists			Course Code:	:	FT215L
Teaching Scheme (Hours)	:	Theory:- 1 hour/week		Total Credits	:	1+1=02	
		Practical:-	2 hours/week				
Evaluation Scheme (Marks)	:	IOE=50 IPE= Nil	EPE= Nil EOE=Nil	Total = 50	Duration of EPE	:	Nil
Revision:	:	Fourth			Month		June 2021
Pre-requisites (if any)	:	Knowledge o	of computer			•	
Types of Course	:	Practical					
Course Domain	:	Core					

Course Rationale: This course exposes students to the depth and breadth of programming practice, with the goal of making students better programmers. It is, however, basic level of course in which some programming concepts are taught.

Course Ol	pjectives		Course Outcomes	
and describe the	er will help to identify e purpose of various the VB integrated onment (IDE).	1.	Students will be able to des build, and debug visual basic ap	• • •
	•	2.	Students will be able to ex Basic's Integrated Denvironment (IDE).	plore Visual Development
	r will help to understand ngs and their use in built	3.	Students will be able to imple rules in visual basic programs.	ment syntax
	er will help to construct petition statements and tures.	4.	Students will be able to explain and data types used in development.	
	er will help to Create ons that store and access les and databases.	5.	Students will be able to apply arithmetic operations for displaying numeric output.	
	ner will help to apply cocedures, and functions ble code.	6.	Students will be able to apply procedures, sub-procedures, and functions to create manageable code.	
	Curriculum Cont	ent		Hours

Unit: I Ir	ntroduction to Visual Basic Programming							
Menu b Controls Controls	par, Tools bar, Project explorer, Tool box, Properties windows, Form Code, s: Command Buttons, Label, Textbox, Pointers, Picture box, frame, Naming s, Properties for controls: Height, Width, Left, Top, Font, Forecolor, Backcolor,	2						
	Caption, Text, and Visible, Events, Saving Visual Basic Project, Examples: Food							
	ring Applications							
Unit: I Mathematics Arithmetic Operations: +, ,*, /, mod, ^. (Using Simple Example for each Operation, Logical Operations. AND, OR, NOT. And the Truth Table for each Operation, (Using Simple Example for each Operation), Relational Operation: >, <, >=, <=, <>, =, String Concatenation (&), Operation Precedence. For all arithmetic, logical, relational operators, Print statement and Formatting, Illustrate (colon, comma, and semicolon), Examples: Food Engineering Applications								
Builtin i	Unit: III Built in Functions Builtin math functions: Abs(x), Int(x), Rnd(x), sgn(x), sqr(x), str(x), val(x), round(x, n), CInt(x), Fix(x), String Functions, Input Box, Msg Box, Examples: Food Engineering Applications.							
Unit: IV Selection Structure/ Reputation Structure Single Selection: If/Then structure, Double Selection: If/Then/Else structure, Nested If/Then/Else structure, Select Case Multiple Selection Structure, For Next Loop, While Wend, Do While Loop, Do Loop Until, Exit Do, Exit For, Examples: Food Engineering Applications								
Unit :V	Variables							
Data Types: Boolean, Integer, Long, Single, Double, String, Valid Naming of Variables, Initial Value for each Type of the Variables (Initial Value for each Data Type), Size of each Variable Type in Bytes, How to Declare Variables. (Dim statement), Using: Dim variable name As Data type, Using Suffix: Integer, Long, Single, Double, String, Constant Variable, Examples: Food Engineering Applications.								
Unit : VI	Database							
	Accessing Database, Data Control DAO, ADO, RDO, Visual Basic and Access Connectivity, 2 Examples: Food Engineering Applications							
Sr. No.	List of Experiments							
1.	To study VB environment							
2.	To design and develop form to perform mathematical operations							
3.	To study date, string and math functions							
4.	To design the form using image control and scroll bars							
5.	To design menu editor as text editor							

6.	To design stop watch
7.	To design form using file controls, OLE control
8.	To design form using Access Database
9-13.	To build a small Food Engineering applications
Sugges	ted Text Books/ Reference Books/Manual
1.	Diane Zak, (2012). Programming with Microsoft Visual Basic. Course Technology, Cengage Learning, 6 th edition.
2.	Evangelos Petroutsos. (2008). Mastering Microsoft Visual Basic. Wiley publishing, Inc., 1^{st} edition.
3.	Noel Jerke. Visual Basic 6: The Complete Reference. Osborne/McGraw-Hill.
4.	Michael Halvorson. E-Book: Microsoft Visual Basic 2010.

Class, Part & Semester	:	Second Year B. Tech (Food Technology), Part II, Semester III								
Course Title	:	E	nvironmental	Course Code	:	HS221				
Teaching Scheme		Lecture : 2	02 hours/we	ek		Total		NI:I		
(Hours)		Tutorial :	Nil			Credits	•	Nil		
Evaluation Scheme (Marks)	:	CIE = Nil SEE = 70	IPE=30 (Project) : Total=100		Duration of SEE	:	3 hrs. At the year end			
Revision	:	Fourth				Month	:	June 2021		
Pre-requisites (if any)	:	Knowledge	Knowledge of Basic Environmental Science							
Course Domain	:	Language and Arts								
Course Rationale: The course is all about learning the way we should live and how we can develop										

Course Rationale: The course is all about learning the way we should live and how we can develop sustainable strategies to protect the environment. It helps individuals to develop an understanding of living and physical environment and how to resolve challenging environmental issues affecting nature.

Course Objectives		Course Outcomes			
The course teacher will explain the types of	1	Students will be able to identify the pollutants and			
environmental pollution.	1.	respond to the pollution problem			
The course teacher will make the students		Students will be able to acquire knowledge of			
recognize social issues and the	2.	ecological threats and choose for sustainable			
environment connectivity with the same.		developments.			
The course teacher will discuss various		Students will be able to anticipate all these laws			
environmental Protection Acts reveal the	3.	and follow the same for the care of the			
students the importance of the same.		environment.			
The course teacher will explain the		Students will be able to apply their knowledge to			
students to adapt to various environmental	4.	implement pollution prevention measure through			
technologies.		some practical work.			
	The course teacher will explain the types of environmental pollution. The course teacher will make the students recognize social issues and the environment connectivity with the same. The course teacher will discuss various environmental Protection Acts reveal the students the importance of the same. The course teacher will explain the students to adapt to various environmental	The course teacher will explain the types of environmental pollution. The course teacher will make the students recognize social issues and the environment connectivity with the same. The course teacher will discuss various environmental Protection Acts reveal the students the importance of the same. The course teacher will explain the students to adapt to various environmental 4.			

Curriculum Content	Hours	
Unit V Environmental pollution: Definition: Causes, effects and control measures of: a) Air		l
pollution, b) Water pollution, c) Soil pollution, d) Marine pollution, e) Noise pollution, f)	ļ	ĺ
Thermal pollution, g) Nuclear hazards Solid waste Management: Causes, effects and control	00	
measures of urban and industrial wastes. Role of an individual in prevention of pollution.	06	l
Pollution case studies Disaster management: Floods, earthquake, cyclone and landslides.		
Tsunami		
Unit VI Social issues and the environment: From Unsustainable to Sustainable development;		l
Urban problems related to energy; Water conservation, rain water harvesting, watershed	00	ĺ
management; Resettlement and rehabilitation of people; its problems and concerns;	08	
Environmental ethics: Issue and possible solutions; Climate change, Global warming, acid rain,		

	ne layer depletion, nuclear accidents and holocaust; Wasteland reclamation; Consumerism							
and	waste products.							
Uni	t VII Environmental protection: Environment Protection Act.; Air (Prevention and Control of							
Pol	ution) Act.; Water (Prevention and control of Pollution) Act; Wildlife Protection Act; Forest							
	servation Act; Population Growth and Human Health, Human Rights. ;Field Work-Visit to a	06						
	al area to document environmental assets river/forest/grassland/hill/mountain or Visit to a							
loca	al polluted site urban/rural/Industrial/Agricultural or Study of common plants, insects, birds							
or Study of simple ecosystems-ponds, river, hill slopes, etc.								
Uni	t VIII Project / Field work:	10						
Sug	gested Text Books:							
1.	Agarwal, K. C. (2001), Environmental Biology, Nidi Publ. Ltd., Bikaner.							
2.	BharuchaErach, The Biodiversity of India, Mapin Publishing Pvt. Ltd., Ahmedabad, 380013, Ind	dia						
3.	Brunner R. C., (1989), Hazardous Waste Incineration, McGraw Hill Inc. 480p.							
Sug	gested Reference Books:							
1.	Clark R. S., Marine Pollution, Clanderson Press Oxford (TB) Pg No. 6							
2.	Cunningham, W. P. Cooper, T. H. Gorhani, E. and Hepworth, M. T. (2001), Environmental							
	Encyclopedia, Jaico Publ. House, Mumbai, 1196p.							
3.	De A. K., Environmental Chemistry, Wiley Eastern Ltd.							
4.	Down to Earth, Centre for Science and Environment (R)							
5.	Gleick, H., (1993), Water in crisis, Pacific Institute for Studies in Dev., Environment &	Security.						
	Stockholm Env. Institute. Oxford Univ. Press 473p.							
6.	Hawkins R. E., Encyclopedia of Indian Natural History, Bombay Natural History Society, Bomba	ay (R).						
7.	Heywood, V. H. and Watson, R. T. (1995), Global Biodiversity Assessment, Cambridge Ur	iv. Press						
•	1140p.							
8.	Jadhav, H. and Bhosale, V. M. (1995), Environmental Protection and Laws, Himalaya Pub Delhi, 284p.	. House,						
9.	Mckinney, M. L. and School. R. M. (1996), Environmental Science Systems and Solution enhanced edition.	ns, Web						
10.	Mhskar A. K., Matter Hazardous, Techno-Science Publications (TB).							
11.	Miller T. G. Jr., Environmental Science, Wadsworth Publishing Co. (TB).							
12.	Odum, E. P. (1971), Fundamentals of Ecology, W. B. Saunders Co. USA, 574p.							
13.	Rao M. N. and Datta, A. K. (1987), Waste Water Treatment, Oxford and IBH Publ. Co. Pvt. Ltd.	,						
14.	Sharma B. K., (2001), Environmental Chemistry, Goel Publ. House, Meerut							
15.	Survey of the Environment, The Hindu (M)							

16.	Townsend C., Harper, J. and Michael Begon, Essentials of Ecology, Blackwell Science (TB).
17.	Trivedi R. K., Handbook of Environmental Laws, Rules, Guidelines, Compliances and Standards, Vol. I
	and II, Enviro Media (R).
18.	Trivedi R. K. and P. K. Goel, Introduction to air pollution Techno-Science Publications (TB)
19.	Wagner K. D., (1998), Environmental Management, W. B. Saunders Co. Philadelphia, USA.

Class, Part & Semester	:	Second '	Second Year B. Tech. (Food Technology), Part II, Semester III								
Course Title	:	Sc	oft :	Course Code	:	HS212					
Teaching Scheme (Hours)	•	2 hours /week				Total Credits	:	Nil			
Evaluation Scheme (Marks)	:	Assignments Viva voce	:	50 25	Written Test Grand Total	:	25 100	Duration of SEE	:	NA	
Revision	••	Fourth	-ourth							June 2021	
Pre-requisites (if any)	:	H. S. C. Level English language competency									
Course Domain	:	Humanity and A	lumanity and Arts								

Course Rationale: The course skills focus on who people are, as opposed to what they are trained in. These skills serve to represent learners' approach to life and work. The course develops interpersonal skills hardwired to an individual's personality, and such skills characterize how we interact with other people in the workplace. These skills are important because they enable students to adjust to the frustrations and challenges they will encounter in their adult life, as well as the demands of work. Mastering soft skills help students learn, live and work better.

Course Assessment Method: The students will be given five assignments each for 10 marks. At the end of the course, there will be a written test of 25 marks and a viva voce of 25 marks. All these assessments will be for a total of 100 marks. Based on the marks obtained, they will be awarded with a grade similar to other credit courses. Though it is an audit course, obtaining passing grade is essential.

	Course Objectives		Course Outcomes
1.	The course teacher will illustrate the components of self-development and state the importance of career planning.	1.	Students will be able to identify components of self-development and realize its importance in their career planning.
2.	The course teacher will define communication and classify the same.	2.	Students will be able to differentiate communication types and apply the same.
3.	The course teacher will explain behavioral skills, team skills and interpersonal skills.	3.	Students will be able to acquire behavioral, team and interpersonal skills and display the same.
4.	The course teacher will classify documentation types and describe various types of report writing.	4.	Students will be able to follow different document formats and acquire report and proposal writing skills.
5.	The course teacher will describe emotional intelligence and its role.	5.	Students will be able to receive and respond to emotions with intelligence.
6.	The course teacher will paraphrase interview skills and demonstrate resume writing.	6.	Students will be able to acquire interview skills and apply those when required.

Curriculum Content	Hours
Unit I Self Development: Self-analysis, creativity, attitude, motivation, goal setting. Importance of career visioning and planning.	02
Unit II Effective Communication Skills: Importance of communication, Communication process, Elements of communication, Communication Types-verbal and non-verbal, objectives of communication. Business Communication, current English usage, debates, language games, situational dialogues, precise writing, essay writing, presentations.	06
Unit III Behavioral Skills: Psychological Tests: Aptitude and personality assessment, suggestions for improvement, Team Skills: Team building and leadership, evolution of groups into teams, group dynamics, emergence of leadership, intra-group dynamics, inter-group dynamics, conflict management, inter dependency, assessment of team-based projects, Time Management: Pareto's Principle, Parkinson's Laws, Murphy's Laws, Law of Clutter, prioritization, goal setting, effective time management, Interpersonal Skills: Negotiations, listening skills, social skills, assertive skills, cross-cultural communications, Leadership Skills: Concepts of leadership, leadership styles, insights from great leaders.	08
Unit IV Documentation: Report writing-Formal report, study tour report, project report, Writing proposal-solicited proposals and unsolicited proposals.	03
Unit V Emotional Intelligence: Emotional Brain, Nature of emotional intelligence, emotional intelligence applied windows of opportunity, emotional literacy.	04
Unit VI Interview Skills: Importance of Interview Skills, Resume Building, Group discussion and personal interview, Psychometric Test, actual career planning.	03
Suggested Text Books:	
1. Soft Skills. (2015). Career Development Centre, Green Pearl Publications.	
Suggested Reference Books:	
1. Covey Sean. (1998). Seven Habits of Highly Effective Teens. New York, Fireside Publishers.	
2. Carnegie Dale. (1998). How to win Friends and Influence People, New York: Simon and Sch	uster.
3. Thomas A Harris (1972). I am ok, You are ok. New York-Harper and Row.	
4. Daniel Goleman. (2006). Emotional Intelligence, Bantam Book.	
5. Effective communication skill, MTD training and Ventus publishing ApS ISBN 978-87-7681-	598-1.

Class, Part & Semester	:	Second Year B. Tech. (Food Technology), Part II, Semester IV						
Course Title	:	Food P	rocess l	Engineering I	Course Code:	:	FT221	
Teaching Scheme (Hours)		Lecture: 3	3 hour	rs/week	Total		0.4	
		Tutorial :1	orial :1 1 hour/week			•	04	
Evaluation Scheme	aluation Scheme : CIE=30		SEE	Grand Total=100	Duration		3 hours	
(Marks)	•	(20+10)	= 70	Grand Total=100	of SEE		3 110013	
Revision:	:	Fourth			Month	:	June 2021	
Pre-requisites	:	Knowledge of	Knowledge of Principles of Food Preservat					
(if any)								
Course Domain	:	Core						

This course deals with mass and energy balances, basic knowledge, principles, working about various mechanical operations and mass transfer operations with their respective applications in the food industry.

	Course Objectives		Course Outcomes
1.	The course teacher will define the course and indicate the importance of the same to the students.	1.	Students will be able to recognize the scope and need of the course.
2.	The course teacher will explain the basic concepts of mass balance and energy balance with its applications in food processing unit operations.	2.	Students will be able tounderstand all unit operations and its applications in food processing.
3.	The course teacher will describe the basic theory and concept involved in food processing unit operations.	3.	Students will be able to apply the principles of mass and energy balance to food processing systems.
4.	The course teacher will enable the students to solve problems in food processing industries.	4.	Students will be able toanalyse the working of the food processing equipment used for the different unit operations.
5.	The course teacher will describe the working of various processing equipment with its application in food processing industries.	5.	Students will be able to evaluate how various unit operations work individually and together.
6.	The course teacher will enable students to design equipment or food processing operations	6.	Students will be able to design and construct the food processing equipment or food processing unit operations.
	Curriculum Co	Hours	

	Introduction and Size Reduction	
	uction: Unit operations in food engineering, Mass and energy balance	7
	eduction: Particle size analysis, equipment, application to food industries	
	Mixing	
	theory of solid – solid mixing, liquid- liquid mixing, equipment, applications to food	6
indust		
	I Filtration and Centrifugation	
	ion :Theory of filtration, industrial filters, applications to food industries	7
	fugation: Theory of centrifugation, equipment, applications to food industries	
	V Distillation	
-	ir liquid equilibria, batch and continuous distillation, steam distillation, equipment,	7
	ations to food industries	
	/ Diffusion and Gas Absorption	
	ion: Basics of mass transfer and diffusion in food systems, Molecular diffusion and	7
	Law; Diffusion of gases in liquid and solid foods	
	osorption: Principles, equipment and applications in food processing	
	/I Extraction	5
Solver	nt extraction, leaching, equipment, applications for food industries	
Sugge	ested Text Books:	
1.	Rao, D. G. (2012). Fundamental of Food Engineering. PHI Learning Private Limite	ed, New
1.	Delhi.	
2.	Singh, R.P., and Heldman, D.R. (2001). Introduction to Food Engineering, 3 rd ed., A	cademic
۷.	Press, San Diego, CA. Academic Press publications. 69–78, 144–157.	
3.	Geankoplis, C. J. (2002). Transport processes and unit operations. 3 rd Edition. Pren	tice Hall
٥.	of India.	
4.	Coulson and Richardsons. (1998). Chemical Engineering , Vol I and II, Asiali Books Pvt I	td.
5.	McCabe and Smith "Unit Operations" McGraw-Hill, New York	
Sugge	ested Reference Books:	
1.	Treybal, R. E. (1981). Mass Transfer Operations. 3 rd edition. McGraw Hill.	
2.	Dennis, R.H. (1981). "Food Process Engineering. Academic Publishing and Press, Ki	ng Saud
	University.	
3.	Rao, M.A. Syed S.H. Rizvi, and Ashim K. Datta. (2008). Engineering properties of foo	ods. CRC
	Press.	
4.	ZekiBer. (2018). Food Process Engineering and Technology. 3 rd Edition. Academic Pres	SS.
-	Sahay, K.M. and Singh, K. K. (2003). UNIT Operations in Agricultural Processin	g. Vikas
5	Publishing House Pvt. Ltd., New Delhi.	
L		

Class, Part & Semester	:	Second Year B. Tech (Food Technology), Part II, Semester IV					
Course Title	:	Food Add	Course Code:	:	FT222		
Teaching Scheme (Hours)		Lecture : 4	4 hours/week	Total Credits	:	04	
		Tutorial : Nil	Nil				
Evaluation Scheme		CIE=30	SEE = 70	Grand Total=	Duration		3 hours
(Marks)	•	(20+10)	3EE = 70	100	of SEE	•	5 HOURS
Revision:	:	Fourth			Month	:	June 2021
Pre-requisites	:	Knowledge of Food Chemistry, Food Microbiology and Principles of Food					
(if any)		Preservation					
Course Domain	:	Core					

Course Rationale: The course covers a broad range of subjects related to foods additive and its practical applications. Introduction to the types of food additives currently used in the food industry and the function(s) of these chemical compounds. The safety of food additives will be addressed as related to their toxicological information. The overarching question of the need for food additives and their safety is being presented. In addition, a selection of naturally-occurring toxicants in foods will be examined and discussed.

Course Objectives		Course Outcomes		
1.	The course teacher will help to understand	1.	Students will be able to describe the	
	the classification of food additives and		various additives and ingredients used in	
	ingredients.		food industries.	
2.	The course teacher will help to understand	2.	Students will be able to describe the	
	the significance of different food additives		mechanisms of food additives involved in	
	and ingredients in food quality,		foods.	
	preservation and storage.			
3.	The course teacher will help to understand	3.	Students will be able to explain the	
the safety of use of food additives and			significance of different additives in food	
	ingredients.		quality, preservation and storage	
4.	The course teacher will help to understand	4.	Students will be able to describe the	
their Maximum Permissible Limit (MPL) of			safety of use of food additives and	
	additives and ingredients in foods.		ingredients.	
5.	The course teacher will help to understand	5.	Students will be able to extrapolate the	
	the effect of different process conditions		knowledge gained on food additives and	
	on stability of food additives and		ingredients in food industries.	
	ingredients.			
6.	The course teacher will help to understand	6.	Students will be able to describe the	
	the process of preparation of food		process of preparation of food additives	
	additives and ingredients.		and ingredients.	

Unit I	Curriculum Content	Hours				
Defin	Definitions of Food Additives, Classification and Functions, Legitimate uses of Additives in					
foods	foods, Intentional and Non Intentional additives, Difference between Additives and					
Adult	erants, Toxicological evaluation of food additives.					
Unit I	I					
	and functions and Applications of: Buffers, Acids, Chelating agents, Low calorie, utritive sweeteners and Polyols, Antioxidants, Antimicrobial Agents	07				
Unit	II					
Uses and functions and Applications of: Emulsifying and stabilizing agents, Anti-caking						
agent	s, Thickeners, Firming agents, Hydrocolloids					
Unit	V					
Use and Functions of Flour bleaching agents and Bread improvers.						
Food Preservatives : Introduction; Classification- Natural and chemical Preservatives; Mode						
of act	ion; Role in Food processing,					
Unit '	V					
Food colorants: Introduction; Natural and Synthetic food colorants; Classification of Food						
	ants; Chemical nature; Impact on health. Application of colors in food.					
	Flavor: Introduction; Classification of food flavor; Application of colors in food.					
Unit '						
	ent toxicants terms in toxicology, safety evaluation using traditional and modern pach contaminants, pesticide residues, toxicology and public health.	10				
Sugg	ested Text Books:					
1.	Furio T. F. (1990) Handback of Food Additives CDC Press Boss Potent Flor					
	Furia, T. E., (1980). Handbook of Food Additives, CRC Press, Boca Raton, Flor.					
2.	Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker.					
2. 3.		ell.				
	Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker.	nization.				
3. 4.	Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker. Msagati, T. A. (2013). The chemistry of food additives and preservatives. Wiley-Blackw Joint, F. A. O., WHO Expert Committee on Food Additives, and World Health Organ (1993). Toxicological evaluation of certain food additives and contaminants. World	nization. d Health				
3.	Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker. Msagati, T. A. (2013). The chemistry of food additives and preservatives. Wiley-Blackw Joint, F. A. O., WHO Expert Committee on Food Additives, and World Health Organ (1993). Toxicological evaluation of certain food additives and contaminants. World Organization.	nization. d Health				
3. 4. 5.	Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker. Msagati, T. A. (2013). The chemistry of food additives and preservatives. Wiley-Blackw Joint, F. A. O., WHO Expert Committee on Food Additives, and World Health Organ (1993). Toxicological evaluation of certain food additives and contaminants. World Organization. Saltmarsh, M. (Ed.). (2020). Saltmarsh's Essential Guide to Food Additives. Royal Science (1993).	nization. d Health				
3. 4. 5.	Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker. Msagati, T. A. (2013). The chemistry of food additives and preservatives. Wiley-Blackw Joint, F. A. O., WHO Expert Committee on Food Additives, and World Health Organ (1993). Toxicological evaluation of certain food additives and contaminants. World Organization. Saltmarsh, M. (Ed.). (2020). Saltmarsh's Essential Guide to Food Additives. Royal Schemistry.	nization. d Health				
3. 4. 5.	Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker. Msagati, T. A. (2013). The chemistry of food additives and preservatives. Wiley-Blackw Joint, F. A. O., WHO Expert Committee on Food Additives, and World Health Organ (1993). Toxicological evaluation of certain food additives and contaminants. World Organization. Saltmarsh, M. (Ed.). (2020). Saltmarsh's Essential Guide to Food Additives. Royal Schemistry.	nization. d Health				
3. 4. 5. Sugge	Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker. Msagati, T. A. (2013). The chemistry of food additives and preservatives. Wiley-Blackw Joint, F. A. O., WHO Expert Committee on Food Additives, and World Health Organ (1993). Toxicological evaluation of certain food additives and contaminants. World Organization. Saltmarsh, M. (Ed.). (2020). Saltmarsh's Essential Guide to Food Additives. Royal Schemistry. Ested Reference Books: Fennema, O. R. (1996). Food Chemistry, Marcel Dekker. New York, NY.	nization. Health ociety of				
3. 4. 5. Sugge 1. 2.	Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker. Msagati, T. A. (2013). The chemistry of food additives and preservatives. Wiley-Blackw Joint, F. A. O., WHO Expert Committee on Food Additives, and World Health Organ (1993). Toxicological evaluation of certain food additives and contaminants. World Organization. Saltmarsh, M. (Ed.). (2020). Saltmarsh's Essential Guide to Food Additives. Royal Sc Chemistry. Ested Reference Books: Fennema, O. R. (1996). Food Chemistry, Marcel Dekker. New York, NY. DeMan. (1982). Food chemistry" Hardcover, A V I Publishing Company.	nization. Health ociety of				
3. 4. 5. Sugge 1. 2.	Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker. Msagati, T. A. (2013). The chemistry of food additives and preservatives. Wiley-Blackw Joint, F. A. O., WHO Expert Committee on Food Additives, and World Health Organ (1993). Toxicological evaluation of certain food additives and contaminants. World Organization. Saltmarsh, M. (Ed.). (2020). Saltmarsh's Essential Guide to Food Additives. Royal Scientistry. Ested Reference Books: Fennema, O. R. (1996). Food Chemistry, Marcel Dekker. New York, NY. DeMan. (1982). Food chemistry" Hardcover, A V I Publishing Company. William C. Frazier And Dennis C. Westoff. (2017). Food Microbiology 5th Edition, McG.	nization. d Health ociety of				

5.	Wood, R., Foster, L., Damant, A., & Key, P. (2004). Analytical methods for food additives.
	Elsevier.
6.	Saltmarsh, M., and Saltmarsh, M. (Eds.). (2013). Essential guide to food additives. Royal
	Society of Chemistry.
7.	Smith, J., and Hong-Shum, L. (2011). Food additives data book. John Wiley and Sons.
8.	Branen, A. L., Davidson, P. M., Salminen, S., and Thorngate, J. (Eds.). (2001). Food additives.
	CRC Press.
9.	Wood, R., Foster, L., Damant, A., & Key, P. (2004). Analytical methods for food additives.
	Elsevier.

Class,	, Part & Semester	:	Second Y	ear B. Te	ch. (F	ood Technology	y), Part II, Se	me	ster	IV
Course Title			Fo	Course Code:	:	I	FT223			
Teaching Scheme			Lecture : 4	4 Hrs/we	eek		Total			04
	(Hours)	•	Tutorial : Nil	Nil			Credits			04
Eva	luation Scheme	•	CIE=30	SEE = 7	70	Grand Total=	Duration		3	hours
	(Marks)	•	(20+10)	JLL - /		100	of SEE	Ŀ		
	Revision:	:	Fourth				Month	:	Ju	ne 2021
P	Pre-requisites	:	Knowledge of b	asic scie	nce					
C	(if any) ourse Domain	_	Coro							
		:	Core							
Cours	se Rationale:			ı						
	Course	Ob	jectives			Cour	rse Outcomes	5		
1.	The course tead		* *		1.		ill be able to	un	ders	tand the
	the understandir	_		•		biochemica	·			
2.	The course teac				2.		vill be able		_	•
	the biosynthesis of macromolecules a			les and		1 -	s of macro			
micro molecules.					techniques.	cules various	100	a pr	ocessing	
3.	The course tead	che	er helps to und	erstand	3.	•				tand the
	the fundamenta		•				fundamental biochemical principles			
	structure/function	n	of bio-mo		structure/fu	inction of	bi	o-m	olecules,	
	metabolic pathw	/ay	s and the regula		metabolic p	athways and	d th	e re	gulation	
	biochemical prod	ces	sses.			of biochemi	cal processes	5.		
4.	The course tead		•		4.		ill be able to	-		
	the mechanism		•				of enzym			
	identify factors a						ors affecting			
5.	The course tead the major che		•		5.		ill be able to emical and			chemical
	1		ons that influence				reactions			
	quality with em				food quality with emphasis on f					
	applications.						industry applications.			
6.	The course tead	che	er helps to und	erstand	6.	Students w	ill be able t	to (desc	ribe the
how the properties of different food						properties o	of different fo	od	com	ponents
components and interactions among these				_				amo	_	these
	components mo		-	quality		•	s modulate			specific
	attributes of foo	d s	systems.			quality attri	butes of food	sys	stem	ıs.
										1
			Curricu	ulum Con	tent					Hours

UNIT-	Digestion, absorption and metabolism	42					
	ion, absorption and metabolism of carbohydrate, fat and protein	12					
UNIT-	I Enzymes						
Chemical nature and nomenclature, classification, sources and properties, mechanism of							
action	, Coenzyme and prosthetic groups application.						
UNIT-	II Biochemical changes in meat	8					
Bioche	emical changes in meat and meat products during processing.	0					
UNIT-	V Biochemical changes in fruits and vegetables						
Bioche	emical changes in fruits and vegetables during post harvest handling, processing and	8					
storag	e.						
	V Biochemical changes in milk	8					
Bioche	emical changes in milk and milk products during handling, processing and storage.						
UNIT-VI Hormones							
Horm	Hormones related to metabolic process						
Sugge	sted Text Books:						
1.	Cox, M.M. and Nelson, David L. Lehininger. (2008). Principles of Biochemistry. 5 th Edition. W.						
1.	H. Freeman.						
2.	Murray, Robert K. et al., (2006). Harper's Illustrated Biochemistry. 27 th Edition. McGraw-Hill.						
3.	Satyanarayanan, U. (2005). Biochemistry. Books and Allied.						
4.	Voet, D. J. and J. G. Voet. (2008). Principles of Biochemistry. 3 rd Edition. John Wiley.						
5.	Rastogi, S.C. (2003). Biochemistry. 2 nd Edition. Tata McGraw-Hill.						
Sugge	sted Reference Books:						
1.	Lehninger. (1993). Biochemistry. CBS. Publications.						
2. Stryer. (1988). Biochemistry. W H Freeman, New York.							
3.	3. LubertStryen. (2006). Biochemistry, 6 th Edition, W H Freeman and Co.						
4.	4. Shubhangini A. Joshi. (2002). Nutrition and Dietetics.2 nd Edition. Tata McGraw Hill publishing						
	Company Ltd.						
5.	J. H. Weil, (1990). General Biochemistry, 6 th Edition, New Age International Private Lim	nited.					

Class, Part & Semester	:	Second	Second Year B. Tech. (Food Technology), Part II, Semester IV							
Course Title	••	Human Nutrition			Course Code:	:	FT224			
Teaching Scheme		Lecture : 4	Lecture : 4 4 hours/week		Total		04			
(Hours)	:	Tutorial : Nil	Nil		Credits	:	04			
Evaluation Scheme		CIE=30	SEE = 70	Grand Total=	Duration of SEE		3 hours			
(Marks)	•	(20+10)	3EE - 70	100		:				
Revision:	:	Fourth	Fourth				June 2021			
Pre-requisites	:	Knowledge of	Knowledge of basic science							
(if any)										
Course Domain	:	Core	Core							

Course Rationale: This course is all about nutritional aspects of foods. The aim of the course is to provide the students with deep understanding of how nutrients and balanced diet contributes to maintain healthy life; and explain how to deal with nutrition related problems, diseases by health promoting balanced diet. Human nutrition deals with physiological and metabolic functions of human digestive system; Nutritional requirements of special group of people such as aged, infants, pregnant & mothers, patients etc. Human nutrition is major aspect of basic nutrients requirements, Balanced diet and awareness about food faddism. It helps to solve nutrient deficiencies diseases, Malnutrition etc. It promotes awareness about RDA, Food components and nutrients that affecting on immune systems such as undesirable constituents, toxic substances and their disorders.

	Course Objectives	Course Outcomes				
1.	The course teacher helps to understand	1.	Students will be able to understand the			
	the physiological and metabolic functions		physiological and metabolic functions of			
	of human digestive system.		nutrients.			
2.	The course teacher helps to understand an	2.	Students will be able to Familiarize			
	overview of the major macro and		nutritional assessment, RDA and Dietary			
	micronutrients relevant to human health.		Recommendations and guidelines.			
3.	The course teacher will help to formulate	3.	Students will be able to understand the			
	dietary recommendations.		importance of energy and water balance.			
4.	The course teacher helps to understand	4.	Students will be able to understand			
	proper diet planning, nutritional facts for		malnutrition, their causes and treatment.			
	balanced nutrition and healthy diets.					
5.	The course teacher helps to understand	5.	Students will be able to understand the			
	the role of diet in causing and preventing		principles involved in the diet, exchange			
	various diseases.		lists, food labels and nutritional facts for			
			balanced nutrition and healthy diets.			
6.	The course teacher helps to get a basic	6.	Students will be able to tell Undesirable			
	foundation in human nutrition in		Constituents and toxic substances and			
	preparation for any of the health		their disorders.			
	professions.					

	Curriculum Content	Hours				
UNIT-	Introduction	8				
Scope	, concepts and importance of nutrition, human digestive system	0				
UNIT-	II Nutritional aspects					
Nutrit	ional aspects of carbohydrate, protein, lipids, water, vitamin and minerals, food, fad	8				
and fa	ddism.					
UNIT-	III Energy and water balance					
Energy	y and water balance, Water intake and losses, energy requirement, and physiological	10				
energy	y value, bomb colorimeter					
UNIT-	IV Malnutrition					
Types	of malnutrition, multi-factorial causes, epidemiology of under nutrition and over	10				
nutriti	on, nutrition infection and immunity, nutrition education					
UNIT- V Balance diet						
Baland	ce diet, types of balance diet, diets for specific purposes.	8				
UNIT- VI Undesirable Constituents and toxic substances						
Undes	irable Constituents and toxic substances and their disorders, hormones	8				
Sugge	sted Text Books:					
1.	Joshi, Shubhangini A., (1992). Nutrition and Dietetics. Tata Mc Grow- Hill publishing C	Company				
	Ltd., New Delhi.					
2.	Geissler. (2009). Fundamentals of Human Nutrition. Elsevier Science.					
3.	Gropper, S. S. (2013). Advance Nutrition and Human Metabolism. Cenage Learning.					
4.	Swaminathan, M. (2006). Advanced Text Book on Food and Nutrition (Volume I and	l II). The				
	Bangalore Printing and Publishing Co. Ltd., Bangalore.					
Sugge	sted Reference Books:					
1.	Stewart Truswell. (2003) .ABC of Nutrition .4 th edition. BMJ Publishing Group.					
2.	Carolyn D. Berdanier, Elaine B. Feldman and Johanna Dwyer. (2008). Handbook of Nutrition					
	and Food. 2 nd Ed. CRC Press, Boca Raton, FL, USA.					
3.	Swaminathan, N. (1987). Food Science and experimental foods. Ganesh Publications, N	1adras.				

Class, Part & Semester	:	Second Year B. Tech (Food Technology), Part II, Semester IV						
Course Title	:	Fluid Mechanics			Course Code:	:	FT225	
Teaching Scheme		Lecture : 4	04 hours/	Total	:	04		
(Hours)	:	Tutorial : Nil	Nil		Credits			
Evaluation Scheme (Marks)	:	CIE=30 (20+10)	SEE = 70	Total=100	Duration of SEE	:	3 hours	
Revision:	:	Fourth	Fourth				June 2021	
Pre-requisites (if any)	:	Knowledge Engineering Mechanics and Engineering physics						
Course Domain	:	Core	Core					

Course Rationale:

This course will give fundamental knowledge of fluid mechanics and use of fluid mechanics in chemical engineering domain.

	Course Objectives	Course Outcome			
1.	The course teacher will explain fundamentals of Fluid Mechanics, which is used in the applications of Aerodynamics, Hydraulics, Marine Engineering, Gas dynamics etc.	1.	Students will be able tostate the Newton's law of viscosity and Explain the mechanics of fluids at rest and in motion by observing the fluid phenomena.		
2.	The course teacher will give fundamental knowledge of fluid, its properties and behavior under various conditions of internal and external flows	2.	Students will be able tocompute force of buoyancy on a partially or fully submerged body and Analyze the stability of a floating body.		
3.	The course teacher willdevelop understanding about hydrostatic law, principle of buoyancy and stability of a floating body and application of mass, momentum and energy equation in fluid flow.	3.	Students will be able toderive Euler's Equation of motion and Deduce Bernoulli's equation.		
4.	The course teacher will help to imbibe basic laws and equations used for analysis of static and dynamic fluids.	4.	Students will be able toexamine energy losses in pipe transitions and sketch energy gradient lines.		
5.	The course teacher will inculcate the importance of fluid flow measurement and its applications in Industries.	5.	Students will be able toevaluate pressure drop in pipe flow using Hagen-Poiseuille's equation for laminar flow in a pipe.		
6.	The course teacher will help to determine the losses in a flow system, flow through	6.	Students will be able todistinguish the types of flows and Determine sonic velocity in a		

pipes, boundary layer flow and flow past immersed bodies.	fluid.				
Curriculum Cor		Hours			
Unit I:Unit systems and Fundamentals to Fluid statics					
Physical quantities, S.I., CGS, FPS engineering units, Conversion of Units, Units and Equations, dimensional analysis, Application of dimensional analysis, Problems. Nature of fluids, Hydrostatic equilibrium, Barometric equation, Hydrostatic equilibrium in centrifugal field, Manometers, Example, U tube, inclined tube manometers.					
Unit II: Fluid flow phenomena and Basic equations of	of fluids flow				
Behavior of flowing fluid, Types of flow, Newtonian momentum flux, viscosities of gases and liquids, viscosity, Flow in boundary layers, Laminar and Tur layer formation in straight tubes, Boundary layer balance, mass velocity, momentum balance, and friction, kinetic energy correction factor, correct equation, Euler's equation, Problems.	Turbulence, Reynolds experiment, Eddy bulent flow in Boundary layers, Boundary r separation and wake formation .Mass Bernoulli's equation without and with tion for fluid friction, Pump Bernoulli's	08			
Unit III: Flow of incompressible fluids in conduits an	d thin layers				
Shear stress distribution in a cylindrical tube, relation friction factor. Relations between skin friction param of Newtonian fluids. Average velocity, kinetic Momentum correction factor (Derivation), Hagen-pound closed channels. Velocity distribution for turb equations for laminar sub layer and buffer layer, Revelocities, Effect of roughness, the friction factor of flow through channels of non-circular section, frictive Effect of fittings and valves, couette flow, Layer flow Problems.	neters .Laminar flow in pipes, Laminar flow energy correction factor (Derivation), piseuille equation. Turbulent flow in pipes ulent flow, universal velocity distribution Relations between maximum and average hart (Moody's diagram), friction factor in ion from changes in velocity or direction,	12			
Unit IV:Transportation and metering of fluids					
Pipe and tubing, joints and fittings. Prevention of Gate valve, globe valve, checks valve butterfly valve, of flowing fluids. Venturimeter, orifice meter, pitot shedding meters, turbine meters, positive displacer meters.	needle valve, ball valve etc. Measurement tube, Rota meters, target meters, vortex	08			
Unit V:Flow past immersed bodies					
Drag coefficients of typical shapes, form drag and st of solids, Ergun's equation, Kozeny- Carman equation Mechanism of fluidization, particulate and aggregical velocity, expansion of -fluidized beds, and application	on, Burke Plummer equation, Fluidization, gative fluidization, minimum fluidization	08			

Unit VI:Flow of compressible fluids and Agitation of fluids

Mach number, continuity equation, Total energy Balance, velocity of sound, ideal gas equations, the asterisk condition, stagnation temperature. Agitation of liquids, Agitation equipment, flow patterns in agitated vessels, circulation rates, Flow numbers, power consumption, power correlations, power correlations for specific impellers, effect of system geometry and calculations for power consumption.

80

Suggested Text Books:

- 1. Mc Cabe, W. L. and Smith, J. C (1993). 'Unit operations of Chemical Engineering.' VII edition, Mcgraw Hill Book Co., International ed.
- 2. Rajput, R. K. Fluid mechanics and Hydraulic machines. VI edition, S. Chand Publication.
- 3. Bansal, R. K. Fluid mechanics and Hydraulic machines.' VII edition, LP Publication.

Suggested Reference Books:

- 1. | Steeter, U.L (1971). Fluid Mechanics' V ed. Mcgraw Hill Book Co., International Edn.
- 2. Richardson, J.E. and Coulson, J.M. C.(1985). Chemical Engineering. 3rd Edition, Vol. 1 Pergamum on Press.
- 3. Miohell, B.I. (1970). Fluid and Particle Mechanics Pergamum on Press.
- 4. Gupta, S.K. (1979). Momentum Transfer Operations, Tata McGraw Hill.

Class, Part & Semester	:	Second Year B. Tech. (Food Technology), Part II, Semester IV						
Course Title	:	Food Process Engineering-I Laboratory			Course Code:	:	FT221L	
Teaching Scheme (Hours)	:	Practical :-	2 hours/week		Total Credits	:	01	
Evaluation Scheme (Marks)	:	IOE= 50 Marks	EOE= 50 Marks	Total= 100 Marks	Duration of EPE		2 hours	
Revision:	:	Fourth	Fourth				June 2021	
Pre-requisites (if any)	:	Knowledge of Principles of Food Preservation						
Course Domain	:	Core	Core					

Course Rationale:

This course deals with mass and energy balances, basic knowledge, principles, working about various mechanical operations and mass transfer operations with their respective applications in the food industry.

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Course Objectives	Course Outcomes

1.	The course teacher will give the knowledge about various unit operations used in food industries.	1.	Students will be able to identify the various unit operations used in food industries.					
2.	The course teacher will describe the basic principles of food processing unit operations.	2.	Students will be able to understand the basic principles of unit operations based on experiment.					
3.	The course teacher will enable student to calculate the mass balance and energy balance equations of food processing operations.	3.	Students will be able to solve the mass balance and energy balance equations of food processes operations.					
4.	The course teacher will explain the working of various equipment involved in various unit operations with experiments	4.	Students will be able to analyze the working of various equipments by practical experience.					
5.	The course teacher will enable the student to evaluate the issues in working of equipment.	5.	Students will be able to evaluate the issues in the working of equipment.					
Sr. No	o. List of Experiments							
1.	Design problems on mass balance and energy	/ bala	nce equations					
2.	Experiment of Particle Size Analysis							
3.	Experiment on Size Reduction							
4.	Experiments on Solid-Solid Mixing							
5.	Experiment on Liquid-Liquid Mixing							
6.	Experiment on Filtration							
7.	Experiment on Centrifugation							
8.	Experiment on sieve tray Distillation							
9.	Experiment on Diffusion							
10.	Experiment on Gas Absorption							
11.	Experiment on Solvent extraction							
12.	Experiment on Leaching							
Sugar	ested Text Books/ Reference Books/Manual							
	Gustavo V Barbosa-Canovas and Li Ma, Blas J Ba	arletta	a. (1997). Food Engineering Laboratory Manual.					
1.	CRC Press.							
2.	2. Christie J. Geankoplis. Transport Processes And Separation Process Principles (Includes Unit Operations) 4 th Edition.							

3. Singh and Heldman. Introduction to food engineering. Academic Press publications

Class, Part & Semester	:	Second Year B. Tech. (Food Technology), Part II, Semester IV						
Course Title	:	Food Additives and Contaminants Laboratory			Course Code:	:	FT 222L	
Teaching Scheme	:	Practical : 2 hours/week		Total		01		
(Hours)		Tractical.	2 Hours, week		Credits	•	01	
Evaluation Scheme		IPE: Nil	EPE: 50	Total= 50	Duration	:	2 hours	
(Marks)	•	IOE: Nil	EOE: Nil	Total= 30	of EPE			
Revision:	:	Fourth			Month		June 2021	
Pre-requisites	:	Knowledge of Food Chemistry, Food Microbiology and Principles of Food					nciples of Food	
(if any)		Preservation						
Course Domain	:	Core						

Course Rationale: The course covers a broad range of subjects related to foods additive and its practical applications. Introduction to the types of food additives currently used in the food industry and the function(s) of these chemical compounds. The safety of food additives will be addressed as related to their toxicological information. The overarching question of the need for food additives and their safety is being presented. In addition, a selection of naturally-occurring toxicants in foods will be examined and discussed.

	Course Objectives		Course Outcomes				
1.	The course teacher will help to train the	1.	Student will get hands on experience				
	students with hands on experience with		with estimation and use of food				
	estimation and use of food additives.		additives.				
2.	The course teacher will help to study the	2.	Students will get knowledge related to				
	basic food additives and their role in food the basic food additives and their ro						
	production.		food production respectively.				
3.	The course teacher will help to study the	3.	Students will learn the effect of food				
	effect of food additives on human health.		additives on human health.				
4.	The course teacher will help to evaluate the	4.	Students will be able to evaluate the				
	quality of developed products.		quality of developed products.				
	•						

Sr. No.	List of Experiments
1.	Detection/Estimation of adulterants in some foods
2.	Determination of carotenoids content
3.	Determination of chlorophyll content
4.	Estimation of tannins content
5.	Extraction of essential oils
6.	Determination of vitamin c content
7.	Effect of acidulants in food products
8.	Effect of thickener in food products

9.	Effect of natural sweeteners/ artificial sweeteners in food products
10.	Effect of stabilizing agents in food products
11.	Effect of emulsifier in food products
12.	Effect of bread improver
Suggest	ed Text Books/ Reference Books/Manual
1.	Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to
1.	follow.
2.	Association of Official Agricultural Chemists and Horwitz, W. (1975). Official methods of analysis
2.	(Vol. 222). Washington, DC: Association of Official Analytical Chemists.
3.	Wood, R., Foster, L., Damant, A., & Key, P. (2004). Analytical methods for food additives. Elsevier.
4.	Saltmarsh, M., and Saltmarsh, M. (Eds.). (2013). Essential guide to food additives. Royal Society of
4.	Chemistry.
5.	Smith, J., and Hong-Shum, L. (2011). Food additives data book. John Wiley and Sons.
6.	Branen, A. L., Davidson, P. M., Salminen, S., &Thorngate, J. (Eds.). (2001). Food additives. CRC
0.	Press.

Class,	Part & Semester	:	Secon	Second Year B. Tech. (Food Technology), Part II, Semester IV					ester IV
Course Title			Food Biochemistry Laboratory			Course Code:	:	FT 223L	
Teaching Scheme : (Hours)			Practical :-	Practical :- 2 hours /week		Total Credits	:	01	
Evalu	uation Scheme (Marks)	:	IPE: Nil IOE: Nil	Total=50			Duration of EPE	:	2 hours
	Revision:	:	Fourth		<u> </u>		Month	:	June 2021
Pr	e-requisites (if any)	:	Knowledge of	Knowledge of Food Chemistry					
Col	urse Domain	:	Core						
Course	Rationale:								
	Course	Ob,	jectives			Со	urse Outcom	es	
1.	The course teacher helps to evaluate the food component on qualitative and quantitative basis.				1.				aluate the food nd quantitative
2.	The course teacher teaches some laboratory techniques and different methodology used in biological assay.			different	2.	The student will be able to know the laboratory techniques and different methodology for biological assay.			
3.		The course teacher helps to conduct qualitative analysis of protein, lipid and fats.			3.	The student will be able to conduct qualitative analysis of protein, lipid and fats.			
4.		The course teacher help to determine vitamin C and vitamin A.			4.		nt will be e techniques f		le to identify analysis.
5.	The course teacher transfer sufficient knowledge of food biochemistry to control reactions in foods.				5.		of food bioch		gain sufficient listry to control
6.	The course teacher helps to develop a practical proficiency in a food biochemistry and analysis laboratory.			•	6.		in a food		elop a practical chemistry and
Sr. No.	List of Experime	ent	s						
1.	Preparation of standard solutions and buffers								
2.	Quantitative determination of carbohydrate by DNSA method								
3.	Isolation and characterization of starch								
4.	Quantitative determination of protein								
5.	Determination of vitamins C								

6.	Determination of vitamins A
7.	Isolation of enzymes from various sources
8.	Enzyme assay methods
9	Immobilization of enzymes
10.	Qualitative analysis of proteins
11.	Qualitative analysis of lipids
12.	Quantitative analysis of lipids (Benedict's method etc.)
13.	Quantitative estimation of blood glucose
Sugges	ted Text Books/ Reference Books/Manual
1.	Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow.

Class, Part & Semester	:	Second Year B. Tech. (Food Technology), Part II, Semester IV					
Course Title	:	Human Nutrition Laboratory			Course Code:	:	FT 224L
Teaching Scheme (Hours)	:	Practical :-	Practical :- 2 hours/week		Total Credits	:	01
Evaluation Scheme (Marks)	:	IPE: Nil IOE: 50	EPE: Nil EOE: Nil	Total=50	Duration of EPE	:	2 hours
Revision:	:	Fourth	Fourth			:	June 2021
Pre-requisites (if any)	:	Knowledge of basic science					
Course Domain	:	Core					

Course Rationale:

The course is all about learning the way we should evaluate BMR Anthropometric measurements, energy value of product. It helps to determine and understanding of micro and macro nutrients such as vitamin A, C, also minerals like iron, Ca, etc. and their requirement for human body. It helps to prepare balanced diet on the basis of physical activity.

	Course Objectives	Course Outcomes		
1.	The course Teacher helps to understand the physiological and metabolic functions of nutrients.	1.	The student will be able to understand the methods used for nutritional assessment.	
2.	The course teacher helps to determine the major macro and micronutrients relevant to human health	2.	The student will be able to familiarize nutritional assessment, RDA and Dietary Recommendations and guidelines.	
3.	The course teacher helps to understand methods of nutritional assessment, RDA and guidelines.	3.	The student will be able to determine BMR and body surface area.	
4.	The course teacher will explain how dietary recommendations are formulated.	4.	The student will be able to understand the food composition and energy balance requires in diet planning.	
5.	The course Teacher will describe the methods used to carry out nutrition research.	5.	The student will be able to understand the methods used to carry out nutrition research.	
6.	The course Teacher helps to compute energy requirement on the basis of physical activity.	6.	The student will able to identify the appropriate techniques requires for Biochemical analysis of blood and urine.	
Sr. No.	List of Experiments			
1.	Calculation of BMR and body surface area			
2.	Calculation of energy value of food			

3.	Preparation of balance diet
4.	Anthropometric measurements
5.	Biochemical analysis of blood
6.	Biochemical analysis of urine
7.	Computation of energy requirement on the basis of physical activity ACU units
8.	Role of various national and international agencies in field of human nutrition
9.	Nutritional labeling of food products
10.	Nutritional survey
11.	Determination of energy value of food by bomb calorimeter
12.	Diet for specific health condition
13.	Visit to Pathological laboratory
Suggest	ed Text Books/ Reference Books/Manual
1.	Swaminathan, M. (2006). Advanced Text Book on Food and Nutrition (Volume I and II). The Bangalore Printing and Publishing Co.Ltd, Bangalore.
2.	Stewart, Truswell. (2003) ABC of Nutrition (4th edition). BMJ Publishing Group. ISBN 0727916645.
3.	Carolyn D. Berdanier, Elaine B. Feldman and Johanna Dwyer. (2008). Handbook of Nutrition and Food. 2nd Ed. CRC Press, Boca Raton, FL, USA.

Class, Part & Semester	:	Second Year B. Tech. (Food Technology), Part II, Semester IV					
Course Title	:	Fluid Mechanics Laboratory			Course Code:	:	FT225L
Teaching Scheme (Hours)	••	Practical :-	02 hours/week		Total Credits	:	01
Evaluation Scheme (Marks)	:	IOE =50	EPE/EOE= Nil Total=50		Duration of EPE	:	Nil
Revision:	:	Fourth	Fourth			:	June 2021
Pre-requisites (if any)	••	Knowledge Engineering Mechanics and Engineering physics					
Course Domain	:	Core	ore				

Course Rationale: Introduction to fluid mechanics laboratory and design of experiments, including experiments on flow patterns, velocity profile in an air pipe, wind tunnel calibration, draining of a tank, pipe friction, drag forces, boundary- layer studies, falling-ball experiments, and measurements of fluid properties.

	Course Objectives		Course Outcomes
1.	The course teacher will help to compare the	1.	Students will be able to compare the
	results of analytical models introduced in		results of analytical models introduced in
	lecture to the actual behavior of real fluid		lecture to the actual behavior of real fluid
	flows.		flows.
2.	The course teacher will discuss standard	2.	Students will be able to use standard
	measurement techniques used in fluid		measurement techniques in fluid
	mechanics and their applications		mechanics.
	respectively.		
3.	The course teacher will help to learn and	3.	Students will be able to write technical
	practice writing the technical reports.		reports.
4.	The course teacher helps to work on small	4.	Students will be able to work on small
	design projects.		design projects.
	·		

Sr. No.	List of Experiments
1.	Study of Bernoulli's Theorem
2.	To Find the Characteristics of Centrifugal Pump
3.	To find out LOSSES in different sizes of Pipe fittings
4.	To determine the friction Losses in Pipes
5.	To determine the coefficient of discharge of Orifice meter
6.	To determine the Velocity of flowing fluid & Coefficient of the given Pitot tube

7.	Measurement of Pressure by using Bourdon Tube Pressure Gauge, Pizometer, U Tube					
	Manometer (500mm)					
8.	8. Measurement of Pressure by using Differential manometer (500 mm & 1000mm), Inclined					
	Manometer (500mm) & Hulshof Apparatus					
9.	To study the performance of Reciprocating Pump and draw characteristic curves.					
10.	To study laminar flow in pipes by using the Reynolds apparatus					
11.	To determine coefficient of discharge of Rotameter					
12.	To determine the Coefficient of Discharge of the given Venturimeter					
Gene	ral Instructions:					
Instit	ute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow.					
Sugg	ested Text Books/ Reference Books/Manual					
1.	Steeter U.L, (1971). Fluid Mechanics. V ed. Mc graw Hill Book Co., International Edn.					
2.	Richardson, J.E. and Coulson, J.M. (1985). Chemical Engineering. 3 rd ed. Vol. 1 Pergamum Press.					
3.	Miohell, B. I. (1970). Fluid and Particle Mechanics Pergamum Press.					
4.	Gupta, S. K. (1979). Momentum Transfer Operations, Tata McGraw Hill.					

Class, Part & Semester	:	Seco	Second Year B. Tech.(Food Technology), Part II, Semester IV					
Course Title	:	Environmental Studies				Course Code	:	HS221
Teaching Scheme		Lecture :	2 hours/week			Total		NI:I
(Hours)	•	Tutorial :	Nil			Credits		Nil
Evaluation Scheme (Marks)	:	CIE = 00 SEE = 70	IPE=30 Project	:	Total=100	Duration of SEE	:	3 hrs. At the year end
Revision	:	Fourth				Month	:	June 2021
Pre-requisites (if any)	:	Knowledge of Basic Environmental Science						
Course Domain	:	Language and Arts						
Course Rationale: The Course is all about learning the way we should live and how we can develop								

Course Rationale: The Course is all about learning the way we should live and how we can develop sustainable strategies to protect the environment. It helps individuals to develop an understanding of living and physical environment and how to resolve challenging environmental issues affecting nature.

	Course Objectives		Course Outcomes			
1	The course teacher will explain the types of	1.	Students will be able to identify the pollutants and			
1.	environmental pollution.	1.	respond to the pollution problem			
	The course teacher will make the students		Students will be able to acquire knowledge of			
2	recognize social issues and the	2.	ecological threats and choose for sustainable			
	environment connectivity with the same.		developments.			
	The course teacher will discuss various		Students will be able to anticipate all these laws			
3.	environmental Protection Acts reveal the	3.	and follow the same for the care of the			
	students the importance of the same.		environment.			
	The course teacher will explain the		Students will be able to apply their knowledge to			
4.	students to adapt to various environmental	4.	implement pollution prevention measure through			
	technologies.		some practical work.			
3.	recognize social issues and the environment connectivity with the same. The course teacher will discuss various environmental Protection Acts reveal the students the importance of the same. The course teacher will explain the students to adapt to various environmental	3.	ecological threats and choose for sustain developments. Students will be able to anticipate all these and follow the same for the care of environment. Students will be able to apply their knowled implement pollution prevention measure the			

Curriculum Content	Hours				
Unit V Environmental pollution: Definition: Causes, effects and control measures of: a) Air		l			
pollution, b) Water pollution, c) Soil pollution, d) Marine pollution, e) Noise pollution, f)	ļ	ĺ			
Thermal pollution, g) Nuclear hazards Solid waste Management: Causes, effects and control					
measures of urban and industrial wastes. Role of an individual in prevention of pollution.					
Pollution case studies Disaster management: Floods, earthquake, cyclone and landslides.					
Tsunami					
Unit VI Social issues and the environment: From Unsustainable to Sustainable development;		l			
Urban problems related to energy; Water conservation, rain water harvesting, watershed	00	ĺ			
management; Resettlement and rehabilitation of people; its problems and concerns;	08				
Environmental ethics: Issue and possible solutions; Climate change, Global warming, acid rain,					

	ne layer depletion, nuclear accidents and holocaust; Wasteland reclamation; Consumerism							
	waste products.							
	Unit VII Environmental protection: Environment Protection Act.; Air (Prevention and Control of							
Poll	ution) Act.; Water (Prevention and control of Pollution) Act; Wildlife Protection Act; Forest							
Con	servation Act; Population Growth and Human Health, Human Rights. ;Field Work-Visit to a	06						
loca	I area to document environmental assets river/forest/grassland/hill/mountain or Visit to a	00						
loca	I polluted site urban/rural/Industrial/Agricultural or Study of common plants, insects, birds							
or S	tudy of simple ecosystems-ponds, river, hill slopes, etc.							
Unit	: VIII Project / Field work:	10						
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Sug	gested Text Books:							
1.	Agarwal, K. C. (2001), Environmental Biology, Nidi Publ. Ltd., Bikaner.							
2.	Bharucha Erach, The Biodiversity of India, Mapin Publishing Pvt. Ltd., Ahmedabad, 380013, In	dia						
3.	Brunner R. C., (1989), Hazardous Waste Incineration, McGraw Hill Inc. 480p.							
Sug	gested Reference Books:							
1.	Clark R. S., Marine Pollution, Clanderson Press Oxford (TB) Pg No. 6.							
2.	Cunningham, W. P. Cooper, T. H. Gorhani, E. and Hepworth, M. T. (2001), Environmental							
۷.	Encyclopedia, Jaico Publ. House, Mumbai, 1196p.							
3.	De A. K., Environmental Chemistry, Wiley Eastern Ltd.							
4.	Down to Earth, Centre for Science and Environment (R).							
5	Gleick, H., (1993), Water in crisis, Pacific Institute for Studies in Dev., Environment & Security.							
;	Stockholm Env. Institute. Oxford Univ. Press 473p.							
6.	Hawkins R. e., Encyclopedia of Indian Natural History, Bombay Natural History Society, Bombay (R)							
7.	Heywood, V. H. and Watson, R. T. (1995), Global Biodiversity Assessment, Cambridge Univ. Press 1140p.							
8.	Jadhav, H. and Bhosale, V. M. (1995), Environmental Protection and Laws, Himalaya Pub. House, Delhi, 284p.							
9.	Mckinney, M. L. and School., R. M. (1996). Environmental Science Systems & Solutions, Web							
10.	Mhskar A. K., Matter Hazardous, Techno-Science Publications (TB).							
11.	Miller T. G. Jr., Environmental Science, Wadsworth Publishing Co. (TB).							
12.	12. Odum, E. P. (1971), Fundamentals of Ecology, W. B. Saunders Co. USA, 574p.							
13.	Rao, M. N. and Datta, A. K. (1987), Waste Water Treatment, Oxford & IBH Publ. Co. Pvt. Ltd.							
14.	Sharma, B. K., (2001), Environmental Chemistry, Goel Publ. House, Meerut							
15.	Survey of the Environment, The Hindu (M)							

16.	Townsend, C. and Harper, J. and Michael Begon, Essentials of Ecology, Blackwell Science (TB).
17.	Trivedi, R. K., Handbook of Environmental Laws, Rules, Guidelines, Compliances and Standards, Vol. I
17.	and II, Enviro Media (R).
18.	Trivedi, R. K. and Goel, P. K. Introduction to air pollution Techno-Science Publications (TB).
19.	Wagner, K. D., (1998), Environmental Management, W. B. Saunders Co. Philadelphia, USA.

Class, Part & Semester	:	Second Year B. Tech.(Food Technology), Part II, Semester IV								
Course Title	:	Introduction to Performing Arts						Course Code	:	HS222
Teaching Scheme (Hours)	:	2 hours /week	2 hours /week					Total Credits	:	Nil
Evaluation Scheme (Marks)	:	Assignments Viva voce	:	50 25	Written Test Grand Total	:	25 100	Duration of SEE	:	NA
Revision	:	Fourth						Month	:	June 2021
Pre-requisites (if any)	:	Student's involvement and interest in the classroom is the pre- requisite.								
Course Domain	:	Humanity and	lumanity and Fine Arts							

Course Rationale:

Course Assessment Method: The students will be given five assignments each for 10 marks. At the end of the course, there will be a written test of 25 marks and a viva voce of 25 marks. There will be assessment for a total of 100 marks. Based on the marks obtained, they will be awarded with a grade similar to other credit courses. Though it is an audit course, obtaining passing grade is essential.

	Course Objectives		Course Outcomes		
1.	The course teacher will state the various performing arts and explain the importance of the same.	1.	Students will be able to identify the types of performing arts and their differences with importance.		
2.	The course teacher will elucidate about drama and Natya-Shastra etc.	2.	Students will be able to acquire knowledge about drama, Natya-Shastra, street play etc.		
3.	The course teacher will explain types of dance, will reveal about theaters.	3.	Students will be able to demonstrate dance skills and organize about theater activities.		
4.	The course teacher will demonstrate about Rag and Taal.	4.	Students will be able to receive and respond to the Rag and Taal.		
5.	The course teacher will explain Gharana system and classify Indian musical instruments.	5.	Students will be able to identify Gharana and instruments of their choice and interest for practice.		
6.	The course teacher will Summarize contribution of great musicians and outline about music concerts	6.	Students will be able to recognize contribution of great musicians and display performances for a music concert.		
	Curriculum Content Hours				

Unit I: Introduction to Music, Dance & Drama, History of Indian Music, Various Forms of Vocal						
Music	Music.					
	II: History and introduction of Drama, Bharat muni natyashastra, street play, Sanskrit	04				
	, Marathi sangitrangbhum					
Unit	III: Dance, its type, greek and roman theatres,	04				
Unit	V: Concept of Raga, Concept of Taal.	04				
Unit	V: Notation System, Study of Gharana system in Music, Classification of Indian	05				
Instru	iments, Instrumental Music.					
	VI: Contribution of Great Musicians, Appreciation of Music. Performance of a Music	05				
Conce	ert.					
Sugg	ested Reference Books:					
1.	. Vasant, Sangeet Karyalaya. Sangeet Visharad. Hatras Prakashan.					
2.	Bidkar, Suchita. (2015). Sangeetshastra Vidnyan Bhag 1, Sanskar Prakashan.					
3.	Bidkar, Suchita. (2015). Sangeetshastra Vidnyan Bhag 2, Sanskar Prakashan.					
4.	Mainkar, Sudhir. Sangeet Kala Aani Shikshan. Sanskar Prakashan.					
5.	Chandavarkar, Bhaskar. (2011). Vadyavedh. 2 nd Edition.SanskarPrakashan.					
6.	6. Mulgaonkar, Arvind. (2015). Tabla. Popular Prakashan.					
7.	7. Chris Hogget. (1977).Stage Crafts. St Martins Pr.					
8.	8. Sarabhai, Mrinalini. Understanding of Bharat Natyam.					
9.	9. Borysenko, Joan. (2007). Minding the body and mending the mind. Bantam.					
10.	10. Subbanna, V.K., Ragadalli Antrang.					

Equivalence of Second Year B. Tech. (Food Technology) Semester III and IV

The Equivalence for the subjects/courses of Food Technology at Second Year B. Tech. Semester III and IV pre-revised Program under the faculty of Engineering and Technology is as follows.

Second Year B. Tech. (Food Technology) Semester III

	Second Year B. Tech.(Food	Second Year B. Tech.(Food	
Sr.	Technology) Semester III	Technology) Semester III	Remarks
No.	Pre-revised syllabus	Revised syllabus	
1.	Principles of Food	Principles of Food	Slight modification in
	Preservation	Preservation	content
2.	Engineering Mathematics-III	Engineering Mathematics-III	Slight modification in content
2	Food Microbiology	Food Microbiology	Slight modification in
3.	Food Microbiology	Food Microbiology	content
4.	Food Chemistry	Food Chemistry	Slight modification in
4.	Food Chemistry	Food Chemistry	content
5.	Heat Transfer	Heat Transfer	Slight modification in
٥.	Treat Transfer	Tieat Hallstei	content
6.	Principles of food	Principles of Food	Slight modification in
	Preservation Laboratory	Preservation Laboratory	content
7.	Food Microbiology	Food Microbiology	Slight modification in
/.	Laboratory	Laboratory	content
8.	Food Chemistry Laboratory	Food Chemistry Laboratory	Slight modification in
0.	Tood Chemistry Laboratory	Tood chemistry Laboratory	content
9.	Heat Transfer Laboratory	Heat Transfer Laboratory	Slight modification in
<i>J</i> .	Treat Transfer Laboratory	Theat Transfer Laboratory	content
10.	Programming practices for	Programming practices for	Slight modification in
10.	Food Technologists	Food Technologists	content
11.	Environmental Studies	Environmental Studies	Slight modification in
11.	Literiorimental studies	Liivii Oiliileittai Studies	content
12.	Soft Skills Development	Soft Skills Davalanment	Slight modification in
12.	Jort James Development	Soft Skills Development	content

Second Year B. Tech. (Food Technology) Semester IV

	Second Year B. Tech. (Food	Second Year B. Tech.(Food	
Sr. No.	Technology) Semester IV	Technology) Semester IV	Remarks
	Pre-revised syllabus	Revised syllabus	
1.	Food Process Engineering-I	Food Process Engineering-I	Slight modification in content
2.	Food Additives and Contaminants	Food Additives and Contaminants	Slight modification in content
3.	Food Biochemistry	Food Biochemistry	Slight modification in content
4.	Human Nutrition	Human Nutrition	Slight modification in content
5.	Fluid Mechanics	Fluid Mechanics	Slight modification in content
6.	Food Process Engineering-I Laboratory	Food Process Engineering-I Laboratory	Slight modification in content
7.	Food Additives and Contaminants Laboratory	Food Additives and Contaminants Laboratory	Slight modification in content
8.	Food Biochemistry Laboratory	Food Biochemistry Laboratory	Slight modification in content
9.	Human Nutrition Laboratory	Human Nutrition Laboratory	Slight modification in content
10.	Fluid Mechanics Laboratory	Fluid Mechanics Laboratory	Slight modification in content
11.	Environmental Studies	Environmental Studies	Slight modification in content
12.	Introduction to Performing Arts	Introduction to Performing Arts	Slight modification in content